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- (54) Composition of polypropylene with propylene-butene elastomer

(57) Disclosed is a polypropylene composition comprising (A) polypropylene and (B) a specific propylene-l-butene random copolymer in a specific amount ratio. The propylene-l-butene random copolymer (B) is characterized in that: (1) the copolymer contains 50 to 95 % by mol of constituent units from propylene and 50 to 5 % by mol of constituent units from 1-butene, (2) the intrinsic viscosity, as measured in decahydronaphthalene at 135 °C, is in the range of 0.1 to 12 dVg, (3) the molecular weight distribution (Mw/Mn), as determined measured by GPC, is not more than 3, and (4) the parameter B value indicating randomness of the copolym-

erized monomer sequence distribution is in the range of 1.0 to 1.5. Also disclosed is a polypropylene composite film comprising (I) a crystalline polypropylene tayer and (II) a tayer formed from the polypropylene composition, which is taminated on at least one surface of the crystalline polypropylene tayer (I). The polypropylene composition is excellent in heat resistance and low-temperature heat-sealing properties as well as in flexibility and impact resistance. The polypropylene composite film is excellent in not only heat-sealing properties, particularly low-temperature heat-sealing properties, but also transparency, scratch resistance and blocking resistance.

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The polypropylene films described above are desired to have further properties such that the films are applicable to high-speed packaging. Thus, the films are desired to have, for example, not only improved low-temperature heat-sealing properties and hot tack properties but also improved slip properties and blocking resistance.

Under such circumstances as mentioned above, the present inventors have studied polypropylene compositions comprising polypropylene and a rubber component, and have found that use of polypropylene and a specific propylene-1-butene random copolymer makes it possible to prepare a polypropylene composition excellent in heat resistance and low-temperature heat-sealing properties as well as in flexibility and impact resistance. The present inventors have also found that a polypropylene film having a heat-sealing layer made of the above-mentioned polypropylene composition has desired properties for applying to the high-speed packaging, such as not only excellent low-temperature heat-sealing properties and hot tack properties but also excellent slip properties and blocking resistance. Moreover, the present inventors have found that the specific propylene-1-butene random copolymer can be prepared using a specific metallocene compound catalyst component. Based on these findings, the present invention has been accomplished.

15 OBJECT OF THE INVENTION

It is an object of the present invention to provide a polypropylene composition which is excellent in heat resistance and low-temperature heat-sealing properties as well as in flexibility and impact resistance.

It is another object of the invention to provide a polypropylene composite film which includes a layer formed from the polypropylene composition and is excellent in not only transparency, low-temperature heat-sealing properties and hot tack properties but also blocking resistance and mechanical properties as scratch resistance.

SUMMARY OF THE INVENTION

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25 According to the present invention, there is provided a polypropylene composition comprising:

- (A) polypropylene in an amount of 5 to 95 % by weight, and
- (B) a propylene-1-butene random copolymer in an amount of 95 to 5 % by weight;
- said propylene-1-butene random copolymer (B) having the following properties:
- (1) the copolymer contains 50 to 95 % by mol of constituent units from propylene and 50 to 5 % by mol of constituent units from 1-butene;
- (2) the intrinsic viscosity, as measured in decahydronaphthalene at 135 °C, is in the range of 0.1 to 12 dVg;
- (3) the molecular weight distribution (Mw/Mn), as determined by get permeation chromatography (GPC), is not more than 3; and
- (4) the parameter B value indicating randomness of the copolymerized monomer sequence distribution is in the range of 1.0 to 1.5, preferably 1.0 to 1.3.

As the polypropylene (A), preferably used is a random copolymer of propylene and an olefin other than propylene, which contains not more than 10 %, by mol of units from the olefin.

In addition to the properties (1) to (4), the propylene-1-butene random copolymer (B) preferably further has the following properties (5) and (6):

(5) the melting point Tm, as measured by a differential scanning calorimeter, is in the range of 60 to 140 °C, and the melting point Tm and the 1-butene constituent unit content M (% by mol) satisfy the following relation:

 $-2.6M + 130 \le Tm \le -2.3M + 155$; and

(6) the crystallinity C, as measured by X-ray diffractometry, and the 1-butene constituent unit content M (% by mol) satisfy the following relation:

C ≥ -1.5M + 75.

The propylene-1-butene random copolymer (B) used in the invention can be prepared using an otefin polymerization catalyst comprising:

(a) a transition metal compound represented by the following formula (1):

$$\begin{array}{c|cccc}
X^1 & X^2 \\
M & & \\
R^3 & R^2 & & \\
R^1 & & \\
R^2 & & R^3 \\
& & \\
R^4 & & \\
\end{array}$$

wherein M is a transition metal of Group IVa, Va or VIa of the periodic table,

R¹ and R² are each a hydrogen atom, a halogen atom, a hydrocarbon group of 1 to 20 carbon atoms, a halogenated hydrocarbon group of 1 to 20 carbon atoms, a silicon-containing group, an oxygen-containing group, a sulfur-containing group, a nitrogen-containing group or a phosphorus-containing group,

(1)

R3 is a hydrocarbon group of 1 to 20 carbon atoms which may be substituted with a halogen atom or a siliconcontaining group,

R4 is a hydrogen atom or an alkyl group of 1 to 20 carbon atoms,

 X^1 and X^2 are each a hydrogen atom, a halogen atom, a hydrocarbon group of 1 to 20 carbon atoms, a halogenated hydrocarbon group of 1 to 20 carbon atoms, an oxygen-containing group or a sulfur-containing group, and

Y is a divalent hydrocarbon group of 1 to 20 carbon atoms, a divalent halogenated hydrocarbon group of 1 to 20 carbon atoms, a divalent silicon-containing group, a divalent germanium-containing group, a divalent tincontaining group, -O-, -CO-, -S-, -SO-, -SO₂-, -NR⁵-, -P(R⁵)-, -P(O) (R⁵)-, -BR⁵- or -AIR⁵- (R⁵ is a hydrogen atom, a halogen atom, a hydrocarbon group of 1 to 20 carbon atoms or a halogenated hydrocarbon group of 1 to 20 carbon atoms);

(b)

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- (b-1) an organoaluminum oxy-compound, and/or
- (b-2) a compound capable of reacting with the transition metal compound (a) to form an ion pair, and, optionally,
- (c) an organoaluminum compound.
- In the present invention, it is preferred to use a transition metal compound (a) of the above formula wherein R¹ is methyl, and also a hydrocarbon group of 2 to 6 carbon atoms.

The present invention further provides a polypropylene composite film comprising:

- (I) a crystalline polypropylene tayer, and
- (II) a layer formed from the polypropylene composition of the present invention, which is laminated on at least one surface of the crystalline polypropylene layer (I).

In the present invention, the polypropylene composition layer (II) is preferably formed from a polypropylene composition comprising:

the polypropylene (A) in an amount of 5 to 50 % by weight, and

the propylene-1-butane random copolymer (B) in an amount of 95 to 50 % by weight; or

the polypropylene (A) in an amount of not less than 50 % by weight and less than 90 % by weight, and

the propylene-1-butene random copolymer (B) in an amount of not less than 10 % by weight and less than 50 % by weight.

The polypropylene (A) is preferably a crystalline polypropylene containing units from propylene in an amount of not less than 90 % by mol.

The intrinsic viscosity (2) of the propylene-1-butene random copolymer (B) is preferably in the range of 0.1 to 5 dVg. The crystalline polypropylene layer [I] may be either unstretched or biaxially stretched.

DETAILED DESCRIPTION OF THE INVENTION

The polypropylene composition of the invention will be described in detail hereinafter.

The polypropylene composition of the invention comprises the following polypropylene and the specific propylene-1-butene random copolymer as described below.

(A) Polypropylene

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In the present invention, any conventionally known polypropylene may be used as the polypropylene (A). The polypropylene may be either a homopolypropylene or a propylene random copolymer containing propylene units and small amounts (e.g., not more than 10 % by mol, preferably less than 5 % by mol) of units from an olefin other than propylene. Of these, the propylene random copolymer is preferably used.

When the polypropylene composition of the invention is intended to be used for a composite film, which will be described later, the polypropylene (A) is preferably a crystalline polypropylene containing units from propylene in amounts of not less than 90 % by mol.

Examples of the other otelins for the propylene random copolymer include α-otelins of 2 to 20 carbon atoms except propylene, such as ethylene, 1-butene, 1-pentene, 1-hexene, 1-heptene, 1-octene, 1-decene, 1-decene, 1-decene, 1-hexendecene and 4-methyl-1-pentene.

In the present invention, it is preferred to employ as the polyproylene (A) a polypropylene prepared by a conventional process using a known solid litenium catalyst component. Also employable is a polypropylene prepared using a metallocene compound catalyst component.

The polypropylene used in the invention has a melting point (Tm) of 100 to 165 °C, preferably 120 to 165 °C. Of the polypropylenes having such melting points, desirably used is a polypropylene having a melting point higher than that of the later-described propylene-1-butene random copolymer.

The polypropylene has a melt flow rate MFR (ASTM D-1238, 230 °C, load of 2.16 kg) of polypropylene in the range of usually 0.1 to 400 g/10 min, preferably 1 to 100 g/10 min, and a molecular weight distribution (Mw/Mn) of more than 3, preferably 4 to 15.

The polypropylene (A) is generally more rigid than the below-described propylene-1-butene random copolymer (B).

(B) Propylene-1-butene random copolymer

The polypropylene composition of the invention comprises the specific propylene-1-butene random copolymer (B) as described below. This propylene-1-butene random copolymer is preferably an elastomeric state.

(1) The propylene-1-butene random copolymer used in the invention is a random copolymer of propylene and 1-butene, and contains:

constituent units from propylene in amounts of 50 to 95 % by mol, preferably 60 to 93 % by mol, more preferably 70 to 90 % by mol, and

constituent units from 1-butene in amounts of 50 to 5 % by mol, preferably 40 to 7 % by mol, more preferably 30 to 10 % by mol.

This propylene-1-butene random copolymer may further contain constituent units from olefins other than propylene and 1-butene, e.g., ethylene, in a small amount, e.g., not more than 10 % by mol.

(2) Intrinsic viscosity (η)

The intrinsic viscosity (η) of the propylene-1-butene random copolymer used in the invention, as measured in decahydronaphthalene at 135 °C, is in the range of 0.1 to 12 d/g, preferably 0.5 to 12 d/g, more preferably 1 to 12 d/g.

(3) Molecular weight distribution

The molecular weight distribution (Mw/Mn) of the propylene-1-butene random copolymer used in the invention, as determined by get permeation chromatography (GPC), is not more than 3, preferably 1.8 to 3.0, more preferably 1.9 to 2.5.

(4) Randomness

The parameter B value of the propylene-1-butene random copolymer used in the invention, which indicates randomness of the copolymerized monomer sequence distribution, is in the range of 1.0 to 1.5, preferably 1.0 to 1.3, more preferably 1.0 to 1.2.

The parameter B value was proposed by B.D. Cole-man and T.G. Fox in J. Polym. Sci., Al. 3188 (1963), and this parameter can be obtained from the following formula:

$$B = P_{12}/(2P_1 \cdot P_2)$$

wherein P₁ is a first monomer content fraction, P₂ is a second monomer content fraction, and P₁₂ is a proportion of the first monomer-second monomer sequences to the total diad sequences.

In the case of B = 1, Bemouth's statistics apply to the copolymer, in the case of B < 1, the copolymer tends to be block; in the case of B > 1, the copolymer tends to be alternating, and in the case of B = 2, the copolymer is an alternating copolymer.

It is preferred that the propylene-1-butene random copolymer (B) used in the invention has, in addition thereto, the above properties (1) to (4) and the following properties (5) and (6).

(5) Melting point Tm

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The melting point Tm of the propylene-1-butene random copolymer (B) used in the invention, as measured by a differential scanning calorimeter, is in the range of 60 to 140 °C, preferably 80 to 130 °C.

It is desired that in the propylene-1-butene random copolymer (B), the melting point Tm and the 1-butene constituent unit content M (% by mot) satisfy the following relation:

(6) It is also desired that in the propylene-1-butene random copolymer (B), the crystallinity C measured by X-ray diffractometry and the 1-butene constituent unit content M (% by mol) satisfy the following relation:

C ≥ -1.5M + 75

The propylene-1-butene random copolymer (B) used in the invention may further have the following property (7).

(7) Stereoregularity of the propylene-1-butene random copolymer (triad facticity, mm fraction)

The stereoregularity of the propylene-1-butene random copolymer used in the invention can be evaluated by triad tacticity (mm (raction).

When head-to-tail enchained three propylene unit sequences in the polymer chains are expressed by a surface zigzag structure, the mm fraction is defined as a proportion of the propylene unit sequences having the same branch directions of the methyl groups and can be determined in the following manner using a ¹³C-NMR spectrum.

For determining the mm fraction of the propylene-1-butene random copolymer from the ¹³C-NMR spectrum, the mm fractions of

- (i) head-to-tail enchained propylene unit triad sequences and
- (ii) propylene unit-butene unit triad sequences consisting of propylene units of head-to-tail enchained propylene units and butene units and containing propylene units as the second units

are measured as the mm fraction of the propylene unit-containing triad sequences present in the polymer chains.

The mm fractions can be determined from the peak intensities of the side chain methyl groups of the second units (propylene units) in the tried sequences (i) and (ii), as described below in detail.

The ¹³C-NMR spectrum of the propylene-1-butene random copolymer is measured for a sample of the random copolymer which is completely dissolved in hexachlorobutadiene containing a small amount of deuterated benzene as a lock solvent by a proton complete decoupling method at 120 °C under the measuring conditions of a flip angle of 45° and a pulse interval of not shorter than 3.4 T₁ (T₁: longest time among the spin-lattice relaxation time of methyl groups). Since T₁ of a methylene group and T₁ of a methine group are shorter than that of a methyl group, the magnetization recovery of all the carbon atoms in the sample is not less than 99 % under the above conditions. As for the chemical shift, the methyl carbon peak of the third unit in the head-to-tail enchained propylene unit pentad sequence (mmmm) is set to 21.593 ppm on the basis of tetramethylsilane, and other carbon peaks are determined based on this peak.

In the ¹³C-NMR spectrum of the propylene-1-butene random copolymer thus measured, the methyl carbon region wherein the side chain methyl groups of the propylene units are observed (about 19.5 to 21.9 ppm) is classified into:

a first peak region (about 21.0 to 21.9 ppm),

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- a second peak region (about 20.2 to 21.0 ppm), and
- a third peak region (about 19.5 to 20.2 ppm).

In each of the above regions, peaks of the side chain methyl groups of the second units (propylene units) in the head-to-tail enchained triad sequences (i) and (ii) are observed as shown in Table 1.

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Shift value	Methyl carbon region (19.5 ~ 21.9 ppm)				
	First region 21.0 ~ 21.9 ppm	Second region 20.2 ~ 21.0 ppm	Third region 19.5 ~ 20.2 ppm		
Hand-to-tail enchainment					
Sequence (i)	PPP (mm)	PPP (mr)	PPP (m)		
Sequence (ii)	PPB (mm)	PPB (mr)			
	BPB (mm)	BPB (mr)			
		PPB (rr)			
		BPB (m)			

In the above table, P indicates a constituent unit from propylene, and B indicates a constituent unit from 1-butene. With regard to the triad sequences (i) consisting of only propylene units, i.e., PPP (mm), PPP (mr) and PPP (rr), among the head-to-tail enchained triad sequences (i) and (ii) shown in Table 1, the directions of the methyl groups are illustrated below by the surface zigzag structures. These PPP sequences apply correspondingly to the mm, mr and rr triad sequences (PPB, BPB) (ii) which contain butene unit.

In the first region, the methyl groups of the second units (propylene units) in the mm triad sequences PPP, PPB and BPB resonate.

In the second region, the methyl groups of the second units (propylene units) in the mr triad sequences PPP, PPB and BPB and the methyl groups of the second units (propylene units) in the m triad sequences PPB and BPB resonate.

In the third region, the methyl groups of the second units (propylene units) in the rr triad sequences PPP resonate.

Accordingly, when (i) the head-to-tail enchained propylene unit triad sequences and (ii) the propylene unit-butene unit triad sequences consisting of head-to-tail enchained propylene units and butene units and containing propylene units as the second units are measured on the side chain methyl groups of the propylene units of the second units in the triad sequences using the ¹³C-NMR spectrum (hexachlorobutadiene solution, tetramethylsilane standard), the triad tacticity (mm fraction) of the propylene-1-butene random copolymor can be determined as a proportion (percentage)

(i) and (ii) are observed, and in the determination of the mm fraction using the above equation, correction is made in the following manner.

The peak area based on the PPE-methyl group can be determined from the peak area of the PPE-methine group (resonance at 30.6 ppm or thereabout), and the peak area based on the EPE-methyl group can be determined from the peak area of the EPE-methine group (resonance at 32.9 ppm or thereabout).

The peak area based on the methyl group C can be determined from the peak area of the adjacent methine group (resonance at 31.3 ppm or thereabout).

The peak area based on the methyl group D can be determined from 1/2 of the total peak area of the peaks based on the αβ methylene carbon in the above structure (iv) (resonance at 34.3 ppm or thereabout and resonance at 34.5 ppm or thereabout), and the peak area based on the methyl group D' can be determined from the area of the peak based on the adjacent methine group (resonance at 33.3 ppm or thereabout) of the methyl group E' in the above structure (v).

The peak area based on the methyl group E can be determined from the peak area of the adjacent methine carbon (resonance at 33.7 ppm or thereabout), and the peak area of the methyl group E' can be determined from the peak area of the adjacent methine carbon (resonance at 33.3 ppm or thereabout).

Accordingly, the peak area of the methyl groups based on the head-to-tail enchained propylene unit triad sequences (i) and (ii) can be determined by subtracting the peak area of the above methyl groups from the total peak area of the peaks in the second and third regions.

Thus, the peak area of the methyl groups based on the head-to-tail enchained propylene unit triad sequences (i) and (ii) can be evaluated, and consequently the mm fraction can be calculated in accordance with the above equation. The carbon peaks in the spectrum can be assigned with reference to a literature "Polymer", 30, 1350 (1989).

(B) Hetero bond unit

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The propylene-1-butene random copolymer (B) used in the invention sometimes has small amounts of structures which contain hetero bond units (regio-irregular units) based on 2,1-insertions or 1,3-insertions of propylene present in the propylene sequences.

In the polymerization, propylene is generally 1,2-inserted (methylene side is bonded to catalyst) to form the aforesaid head-to-tail enchained propylene sequence, but in rare cases propylene is 2,1-inserted or 1,3-inserted. The 2,1-inserted propylenes or the 1,3-inserted propylenes form regio-irregular units represented by the aforementioned structures (iii), (iv) and (v) in the polymer. The proportions of the 2,1-insertions and the 1,3-insertions of propylene in the polymer constituent units can be determined from the following equation using a ¹³C-NMR spectrum with reference to "Polymer", 30, 1350 (1989), similarly to the aforesaid triad facticity.

The proportion of the regio-irregular units based on the 2,1-insertions of propylene can be determined from the following equation.

- -- When it is difficult to directly determine the area of loβ or the like from the spectrum-because of, for example, overlapping of the peaks, correction can be made using the carbon peaks having corresponding areas.

In the propylene-1-butene random copolymer (B) used in the invention, the hetero bond units based on the 2,1-insertions of propylene present in the propylene sequences may be contained in an amount of not less than 0.01 %, specifically 0.01 to 0.3 %, based on all the propylene units.

The regio-irregular units based on the 1,3-insertions of propylene may be contained in the propylene-1-butene random copolymer (B) in an amount of not more than 0.05 %, and this proportion of the regio-irregular units based on the 1,3-insertions of propylene can be determined from the $\beta\gamma$ peak (resonance at 27.4 ppm or thereabout).

Such a propylene-1-butene random copolymer as mentioned above, which is used in the invention, contains a tacky ingredient in a smaller amount than the rubbers conventionally known.

The propylene-1-buttene random copolymer (B) used in the invention can be prepared by copolymerizing propylene and 1-buttene in the presence of an ofelin polymerization catalyst comprising:

(a) a transition metal compound described later;

(b)

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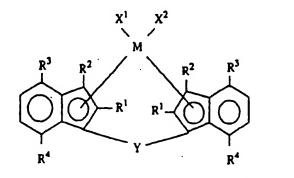
- (b-1) an organoaluminum oxy-compound, and/or
- (b-2) a compound capable of reacting with the transition metal compound (a) to form an ion pair;

and, optionally,

(c) an organoaluminum compound.

Next, the olefin polymerization catalyst used for preparing the propylene-1-butene random copolymer (B) in the invention is described in detail

The transition metal compound (a) (hereinafter sometimes referred to as "component (a)") for forming the olefin polymerization catalyst used for the invention is represented by the following formula (1):



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wherein M is a transition metal of Group IVa, Group Va or Group VIa of the periodic table. Examples thereof include titanium, zirconium, halnium, vanadium, niobium, tantalum, chromium, molybdenum and tungsten. Of these, preferred are titanium, zirconium and hafnium. Particularly preferred is zirconium.

(1)

30 Substituents R1 and R2

R¹ and R² are each independently a hydrogen atom, a halogen atom, a hydrocarbon group of 1 to 20 carbon atoms, a halogenated hydrocarbon group of 1 to 20 carbon atoms, a silicon-containing group, an oxygen-containing group, a sulfur-containing group, a nitrogen-containing group or a phosphorus-containing group.

Examples of the halogens include fluorine, chlorine, bromine and iodine.

Examples of the hydrocarbon groups of 1 to 20 carbon atoms include alkyl groups, such as methyl, ethyl, propyl, butyl, hexyl, cyclohexyl, octyl, nonyl, dodecyl, eicosyl, norbornyl and adamantyl; alkenyl groups, such as vinyl, propenyl and cyclohexenyl; anylalkyl groups, such as benzyl, phenylethyl and phenylpropyl; and anyl groups, such as phenyl, tolyl, dimethylphenyl, trimethylphenyl, ethylphenyl, propylphenyl, biphenyl, naphthyl, methylnaphthyl, anthracenyl and phenanthryl.

Examples of the halogenated hydrocarbon groups include those obtained by substituting the above-mentioned hydrocarbon groups with halogen atoms.

Examples of the silicon-containing groups include monohydrocarbon-substituted silyts, such as methylsilyl and phenylsilyl; dihydrocarbon-substituted silyts, such as dimethylsilyl and diphenylsilyl; trihydrocarbon-substituted silyts, such as trimethylsilyl, triephylsilyl, tricyclohexylsilyl, triphenylsilyl, dimethylphenylsilyl, methyldiphenylsilyl, tritolylsilyl and trinaphthylsilyl; silyl ethers of hydrocarbon-substituted silyts, such as trimethylsilyl ether, silicon-substituted alkyl groups, such as trimethylsilylmethyl, and silicon-substituted aryl groups, such as trimethylsilylmethyl.

Examples of the oxygen-containing groups include hydroxyl groups; alkoxy groups, such as methoxy, ethoxy, propoxy and butoxy; anyloxy groups, such as phenoxy, methylphenoxy, dimethylphenoxy and naphthoxy, and anylalkoxy groups, such as phenylmethoxy and phenylethoxy.

Examples of the sulfur-containing groups include those wherein oxygen is replaced with sulfur in the above-exemplified oxygen-containing groups.

Examples of the nitrogen-containing groups include amino groups; alkylamino groups, such as methylamino, dimethylamino, diethylamino, dipropylamino, dibutylamino and dicyclohexylamino; and arylamino groups and alkylarylamino groups, such as phenylamino, diphenylamino, diolylamino, dinaphthylamino and methylphenylamino.

Examples of the phosphorus-containing groups include phosphino groups, such as dimethylphosphino and diphenylphosphino.

Of these, R1 is preferably hydrogen, methyl, a hydrocarbon group of 2 to 6 carbon atoms and an aromatic group,

and particularly preferably methyl and a hydrocarbon group of 2 to 6 carbon atoms.

P2 is preferably hydrogen and a hydrocarbon group, and particularly preferably hydrogen.

Substituent P3

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R3 is a hydrocarbon group of 1 to 20 carbon atoms which may be substituted with a halogen atom or a siliconcontaining group, and is preferably a secondary or tertiary alkyl group of 3 to 20 carbon atoms or an aromatic group.

Examples of the secondary or tertiary alkyl groups include i-propyl, i-butyl, sec-butyl, tert-butyl, 1,2-dimethylpropyl, 2,3-dimethylbutyl, iso-pentyl, tert-pentyl, neopentyl, cyclohexyl, 4-methylcyclohexyl, iso-hexyl, norbornyl and adamantyl.

Examples of the aromatic groups include aryl groups, such as phenyl, tolyl, dimethytphenylyl, trimethylphenyl, ethylphenyl, propylphenyl, biphenyl, α- or β-naphthyl, methylnaphthyl, anthracenyl, phenanthryl, benzylphenyl, pyrenyl, acenaphthyl, phenalenyl, aceanthrylenyl, tetrahydronaphthyl, indanyl and biphenylyl; and arylafkyl groups, such as benzyl, phenylethylyl, phenylpropyl and tolylmethyl. These groups may contain double or triple bonds.

Further, these groups may be substituted with halogen atoms and silicon-containing groups as described for R1.

Substituent P4

R4 is a hydrogen atom or an alkyl group of 1 to 20 carbon atoms.

Examples of the alkyl groups include chain and cyclic alkyl groups, such as methyl, ethyl, n-propyl, i-propyl, nbutyl, i-butyl, sec-butyl, tert-butyl, pentyl, hexyl, cyclohexyl, heptyl, octyl, nonyl, dodecyl, eicosyl, norbornyl and adamantyl.

These groups may be substituted with halogen atoms and silicon-containing groups as described for R1.

5 X1 and X2

 X^1 and X^2 are each a hydrogen, atom, a halogen atom, a hydrocarbon group of 1 to 20 carbon atoms, a halogenated hydrocarbon group of 1 to 20 carbon atoms, an oxygen-containing group or a sultur-containing group.

Examples of the halogen atoms, the oxygen-containing groups, the hydrocarbon groups of 1 to 20 carbon atoms and the halogenated hydrocarbon groups of 1 to 20 carbon atoms are those described for R¹.

Examples of the sulfur-containing groups include those described for R1; and further sulfonato groups, such as methylsulfonato, trifluoromethanesulfonato, phenylsulfonato, benzylsulfonato, p-toluenesulfonato, trimethylbenzenesulfonato, triisobuty benzenesulfonato, p-chlorobenzenesulfonato and pentafluorobenzenesulfonato; and sulfinato groups, such as methylsulfinato, phenylsulfinato, benzenesulfinato, p-toluenesulfinato, trimethylbenzenesulfinato and pentafluorobenzenesulfinato.

Y

Y is a divalent hydrocarbon group of 1 to 20 carbon atoms, a divalent halogenated hydrocarbon group of 1 to 20 carbon atoms, a divalent silicon-containing group, a divalent germanium-containing group, a divalent tin-containing group, -O-, -CO-, -S-, SO-, -SO₂-, -NR5-, -P(R5)-, -P(O) (R5)-, -BR5- or -AIR5- (R5 is a hydrogen atom, a halogen atom, a hydrocarbon group of 1 to 20 carbon atoms).

More specifically, there can be exemplified:

The divalent hydrocarbon groups of 1 to 20 carbon atoms include alkylene groups, such as methylene, dimethyl-methylene, 1,2-ethylene, dimethyl-1,2-ethylene, 1,3-trimethylene, 1,4-tetramethylene, 1,2-cyclohexylene and 1,4-cyclohexylene, and arylalkylene groups, such as diphenylmethylene and diphenyl-1,2-ethylene.

The divalent halogenated hydrocarbon groups of 1 to 20 carbon atoms include the above-exemplified divalent hydrocarbon groups, which are halogenated, such as chloromethylene;

The divalent silicon-containing groups include alkylsilylene, alkylarylsilylene and anylsilylene groups, such as methylsilylene, diethylsilylene, di(n-propyl)silylene, di(i-propyl)silylene, di(cyclohexyl)silylene, methylphenylsilylene, di(p-tolyl)silylene and di(p-chlorophenyl)silylene, and alkyldisilylene, alkylaryldisilylene and aryldisilylene groups, scuh as tetramethyl-1,2-disilylene and tetraphenyl-1,2-disilylene.

The divalent germanium-containing groups obtained include those wherein silicon is replaced with germanium in the above-exemplified divalent silicon-containing groups.

The divalent tin-containing groups include those wherein silicon is replaced with tin in the above-exemplified divalent silicon-containing groups.

Examples of R5 are halogen atoms, hydrocarbon groups of 1 to 20 carbon atoms and halogenated hydrocarbon groups of 1 to 20 carbon atoms as described for R1.

```
rac-Dimethylsilylene-bis[1-(4-phenylindenyl)]zirconium dichloride,
            rac-Dimethylsilylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium dichloride,
            rac-Dimethylsilylene-bis{1-(2-methyl-4-(α-naphthyl))indenyl))zirconium dichloride.
            rac-Dimethylsilytene-bis[1-(2-methyl-4-(β-naphthyl))indenyl))zirconium dichloride.
            rac-Dimethylsilylene-bis{1-(2-methyl-4-(1-anthracenyl)indenyl)}zirconium dichloride,
            rac-Dimethylsilylene-bis[1-(2-methyl-4-(2-anthracenyl)indenyl)]zirconium dichloride,
            rac-Dimethylsilylene-bis{1-(2-methyl-4-(9-anthracenyl)indenyl)}zirconium dichloride,
            rac-Dimethylsilylene-bis{1-(2-methyl-4-(9-phenanthryl)indenyl))zirconium dichloride,
            rac-Dimethylsilylene-bis{1-(2-methyl-4-(p-fluorophenyl)indenyl)}zirconium dichloride,
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            rac-Dimethylsilylene-bis{1-(2-methyl-4-(pentafluorophenyl))indenyl)}zirconium dichloride,
            rac-Dimethylsilylene-bis[1-(2-methyl-4-(p-chlorophenyl)-1-indenyl)]zirconium dichloride,
            rac-Dimethylsilylene-bis{1-(2-methyl-4-(m-chlorophenyl)indenyl)}zirconium dichloride,
            rac-Dimethylsilylene-bis{1-(2-methyl-4-(o-chlorophenyl)indenyl)}zirconium dichloride,
           rac-Dimethylsilylene-bis[1-{2-methyl-4-(o,p-dichlorophenyl)indenyl)}zirconium dichloride,
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           rac-Dimethylsilylene-bis[1-(2-methyl-4-(p-bromophenyl))indenyl))zirconium dichloride,
           rac-Dimethylsitylene-bis{1-(2-methyl-4-(p-totyl)indenyl)]zirconium dichloride,
           rac-Dimethylsitylene-bis{1-(2-methyl-4-(m-tolyl)indenyl){zirconium dichloride.
           rac-Dimethylsilylene-bis{1-(2-methyl-4-(0-tolyl)indenyl)}zirconium dichloride,
           rac-Dimethylsilylene-bis[1-(2-methyl-4-(0,0'-dimethylphenyl)indenyl))zirconium dichloride,
           rac-Dimethylsilylene-bis[1-(2-methyl-4-(p-ethylphenyl)indenyl)]zirconium dichloride,
           rac-Dimethylsilylene-bis{1-(2-methyl-4-(p-i-propylphenyl))indenyl)}zirconium dichloride,
           rac-Dimethylsilylene-bis{1-(2-methyl-4-(p-benzylphenyl))ndenyl)}zirconium dichloride,
           rac-Dimethylsilylene-bis[1-(2-methyl-4-(p-biphenylyl)indenyl)]zirconium dichloride,
           rac-Dimethylsitylene-bis{1-(2-methyl-4-(m-biphenylyl)indenyl))zirconium dichloride,
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           rac-Dimethylsilylene-bis[1-(2-methyl-4-(p-trimethylsilylenephenyl)indenyl)}zirconium dichloride,
           rac-Dimethylsilylene-bis[1-(2-methyl-4-(m-trimethylsilylenephenyl)indenyl)]zirconium dichloride,
           rac-Dimethylsilylene-bis{1-(2-phenyl-4-phenylindenyl)}zirconium dichloride,
           rac-Diethylsilylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride.
           rac-Di(i-propyl)silylene-bis{1-{2-methyl-4-phenylindenyl)}zirconium dichloride,
           rac-Di(n-butyl)silylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride.
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           rac-Dicyclohexylsilylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride,
           rac-Methylphenylsilylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride,
           rac-Diphenylsilylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride,
           rac-Di(p-tolyt)silylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium dichloride,
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           rac-Di(p-chlorophenyl)silylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride,
           rac-Methylene-bis(1-(2-methyl-4-phenylindenyl))zirconium dichloride.
           rac-Ethylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium dichloride.
           rac-Dimethylgermytene-bis[1-(2-methyl-4-phenylindenyl)]zirconium dichloride,
           rac-Dimethylstannylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride,
           rac-Dimethylsilylene-bis{1-(2-methyl-4-phenylindenyl))zirconium dibromide,
           rac-Dimethylsitylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium dimethyl,
           rac-Dimethylsitylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium methylchloride,
           rac-Dimethylsilylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium chloride SO<sub>2</sub>Me,
           rac-Dimethylsitylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium chloride OSO2Me,
           rac-Dimethylsilylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium mono(trittuoromethanesultonate),
45
          rac-Dimethylsilylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium di(trifluoromethanesulfonate),
          rac-Dimethylsitylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium di(p-toluenesullonate),
          rac-Dimethylsilylene-bis(1-(2-methyl-4-phenylindenyl))zirconium di(methylsullonate),
          rac-Dimethylesilylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium di(trifluoroacetate),
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          rac-Dimethylsilylene-bis[1-(2-methyl-4-phenylindenyl)]zirconium monochloride mono(n-butoxyde),
          rac-Dimethylsilylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium di (n-butoxyde).
          rac-Dimethylsilylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium monochloride mono(phenoxyde),
          rac-Dimethylsilylenc-bis[1-(2-methyl-4-phenylindenyl)]titanium dichloride, and
          rac-Dimethylsilylene-bis[1-(2-methyl-4-phenylindenyl)]halnium dichloride.
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Of the above compounds, particularly preferred are compounds of the formula (1-a) wherein R1 is methyl.

Also preferably used are transition metal compounds of the formula (1-a) wherein R1 is a hydrocarbon group of 2 to 6 carbon atoms and R3 is an aryl group of 6 to 16 carbon atoms. Examples of such compounds are listed below.

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rac-Dimethylsilylene-bis{1-(2-i-butyl-4-(2-methyl-1-naphthyl)indenyl)}zirconium dichloride,
           rac-Dimethylsilylene-bis{1-(2-i-butyl-4-(5-acenaphthyl)indenyl)}zirconium dichloride.
           rac-Dimethylsilylene-bis[1-(2-i-butyl-4-(9-anthracenyl)indenyl)]zirconium dichloride.
           rac-Dimethylsitytene-bis{1-(2-i-butyl-4-(9-phenanthryf)indenyl)}zirconium dichloride.
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           rac-Dimethylsilylene-bis[1-(2-neopentyl-4-phenylindenyl)]zirconium dichloride,
           rac-Dimethylsilylene-bis[1-(2-neopentyl-4-(α-naphthyl)indenyl)]zirconium dichloride,
           rac-Dimethylsilylene-bis[1-(2-n-hexyl-4-phenylindenyl)]zirconium dichloride,
           rac-Dimethylsilylene-bis[1-(2-n-hexyl-4-(a-naphthyl)indenyl)]zirconium dichloride,
           rac-Methylphenylsilylene-bis[1-(2-ethyl-4-phenylindenyl)]zirconium dichloride,
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           rac-Methylphenylsitylene-bis{1-(2-ethyl-4-(α-naphthyl)indenyl)}zirconium dichloride,
           rac-Methylphenylsilylene-bis[1-(2-ethyl-4-(9-anthracenyl))ndenyl)]zirconium dichloride,
           rac-Methylphenylsitylene-bis[1-(2-ethyl-4-(9-phenanthryl)indenyl)]zirconium dichloride,
           rac-Diphenylsitylene-bis[1-(2-ethyl-4-phenylindenyl)]zirconium dichloride,
           rac-Diphenylsitylene-bis(1-(2-ethyl-4-(α-naphthyl))indenyl))zirconium dichloride.
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           rac-Diphenylsilylene-bis(1-(2-ethyl-4-(9-anthracenyl)indenyl))zirconium dichloride,
           rac-Diphenylsilylene-bis[1-(2-ethyl-4-(9-phenanthryl)indenyl)]zirconium dichloride,
           rac-Diphenylsitylene-bis[1-(2-ethyl-4-(4-biphenylyl)indenyl)]zirconium dichloride.
           rac-Methylene-bis[1-(2-ethyl-4-phenylindenyl)]zirconium dichloride,
          rac-Methylene-bis{1-(2-ethyl-4-(α-naphthyl)indenyl)}zirconium dichloride,
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           rac-Ethylene-bis[1-(2-ethyl-4-phenylindenyl)]zirconium dichloride,
          rac-Ethylene-bis[1-(2-ethyl-4-(α-naphthyl)indenyl))zirconium dichloride,
          rac-Ethylene-bis[1-(2-n-propyl-4-(a-naphthyl))indenyl)]zirconium dichloride,
          rac-Dimethylgermylene-bis[1-(2-ethyl-4-phenylindenyl)]zirconium dichloride,
          rac-Dimethylgermylene-bis{1-(2-ethyl-4-(α-naphthyl)indenyl))zirconium dichloride,
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          rac-Dimethylgermylene-bis[1-(2-n-propyl-4-phenylindenyl)]zirconium dichloride,
          rac-Dimethylstannylene-bis(1-(2-ethyl-4-phenylindenyl))zirconium dichloride,
          rac-Dimethyl stannylene-bis \{1-\{2-ethyl-4-\{\alpha-naphthyl\}\} indenyl\}\} zirconium\ dichloride,
          rac-Dimethylstannylene-bis{1-{2-n-ethyl-4-{9-phenanthryl})indenyl}}zirconium dichloride, and
          rac-Dimethylstannylene-bis[1-(2-n-propyl-4-phenylindenyl)]zirconium dichloride.
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In the present invention, also employable are transition metal compounds wherein zirconium metal is replaced with titanium, halnium, vanadium, niobium, tantalum, chromium, molybdenum or tungsten in the above-exemplified compounds.

The transition metal compounds are generally used in the form of racemic modification as the olefin polymerization catalyst component, but they can also be used either in the form of R type or S type.

The transition metal compounds used for the invention can be prepared in accordance with "Journal of Organo-metallic Chem.", 288 (1985), pp. 63-67 and EP-A-0320762. For example, the compounds of the formula (1-a) can be prepared by the following reaction route.

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wherein Z is CI, Br, I or o-tosyl group, and H2Ra is

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$$\bigoplus_{H}^{R^1}$$

The organical uminum oxy-compound (b-1) (hereinafter sometimes referred to as "component (b-1)") for forming the clefin polymerization catalyst used in the invention may be either aluminoxane conventionally known or a benzeneinsoluble organical uminum oxy-compound, the latter is described in, for example, Japanese Patent Laid-Open Publication No. 78687/1990.

The conventionally known aluminoxane can be prepared by, for example, the following procedures.

- (1) An organoaluminum compound, e.g., trialkylaluminum, is added to a hydrocarbon medium suspension of compounds containing adsorbed water or salts containing water of crystallization, e.g., magnesium chloride hydrate, copper sulfate hydrate, aluminum sulfate hydrate, nickel sulfate hydrate and cerous chloride hydrate, so as to allow the organoaluminum compound to react with the adsorbed water or the water of crystallization.
- (2) Water, ice or water vapor is allowed to directly act on an organicaluminum compound, e.g., trialkytaluminum, in a medium such as benzene, toluene, ethyl ether or tetrahydrofuran.
- (3) An organotin oxide such as dimethyltin oxide or dibutyltin oxide is allowed to react with an organoaluminum compound, e.g., trialkylatuminum, in a medium such as decane, benzene or toluene.

The aluminoxane may contain a small amount of an organometallic component. Solutions of the aluminoxane obtained above may be used in the invention, but it is also possible that the solvent or the unreacted organoaluminum compound is distilled off from the solutions and then the residue is redissolved in a solvent or suspended in a poor solvent for aluminoxane.

Examples of the organial uninum compounds used for preparing the aluminoxane include:

trialkylaluminums, such as trimathylaluminum, triethylaluminum, triipropylaluminum, triisopropylaluminum, triisoputylaluminum, triisoputylaluminum, triisoputylaluminum, triisoputylaluminum, triisoputylaluminum, triisoputylaluminum, triisoputylaluminum, trioctylaluminum and tridacylaluminum;

tricycloalkytaluminums, such as tricyclohexytaluminum and tricyclooctytaluminum; dialkytaluminum halides, such as dimethytaluminum chloride, diethytaluminum chloride, diethytaluminum chloride; and diisobutytaluminum chloride;

dialkylaluminum hydrides, such as diethylaluminum hydride and diisobutylaluminum hydride;

dialkylaluminum alkoxides, such as dimethylaluminum methoxide and diethylaluminum ethoxide; and dialkylaluminum aryloxides, such as diethylaluminum phenoxide.

Of these, preferred are trialkylaluminums and tricycloalkylaluminums, and particularly preferred is trimethylaluminum.

Also employable as the organical uninum compound used for preparing the aluminoxane are alkenylaluminums represented by the following formula:

wherein x, y and z are each a positive number, and $z \ge 2x$, such as isoprenylaluminum.

The organization compounds mentioned above are used singly or in combination.

Examples of the solvents used for preparing solutions and suspensions of the aluminoxane include aromatic hydrocarbons, such as benzene, toluene, xylene, cumene and cymene, aliphatic hydrocarbons, such as pentane, hexane, heptane, octane, decane, dodecane, hexadecane and octadecane; alicyclic hydrocarbons, such as cyclopentane, cyclohexane, cyclopentane and methylcyclopentane; petroleum fractions, such as gasoline, kerosine and gas oil; and halogenated derivatives of these aromatic, aliphatic and alicyclic hydrocarbons, particularly chlorides and bromides thereof. Also employable are ethers such as ethyl ether and tetrahydrofuran. Of these solvents, preferred are aromatic hydrocarbons and aliphatic hydrocarbons.

The compound (b-2) capable of reacting with the transition metal compound (a) to form an ion pair (hereinafter sometimes referred to as "component (b-2)"), that is used for forming the ofelin polymerization catalyst according to the invention, includes Lewis acids, ionic compounds and carborane compounds, as described in, for example, Japanese Patent Laid-Open Publications No. 501950/1989, No. 502036/1989, No. 179005/1991, No. 179006/1991, No. 207703/1991 and No. 207704/1991, and U.S. Patent No. 547,718.

Examples of the Lewis acids include triphenylboron, tris(4-fluorophenyl)boron, tris(p-tolyl)boron, tris(0-tolyl)boron, tris(3,5-dimethylphenyl)boron, tris(pentafluorophenyl)boron, MgCl₂, Al₂O₃ and SiO₂-Al₂O₃.

Examples of the ionic compounds include triphenylcarbeniumtetrakis(pentafluorophenyl)borate, tri-n-butylammoniumtetrakis(pentafluorophenyl)borate, N,N-dimethylaniliniumtetrakis(pentafluorophenyl)borate and ferroceniumtetrakis(pentafluorophenyl)borate.

Examples of the carborane compounds include dodecaborane, 1-carbaundecaborane, bis-n-butylammonium (1-carbadodeca)borate, tri-n-butylammonium(7,8-dicarbaundeca)borate and tri-n-butylammonium(tridecahydride-7-carbaundeca)borate.

The compounds (b-2) capable of reacting with the transition metal compound (a) to form an ion pair can be used alone or in combination.

The organoaluminum compound (c) (hereinafter sometimes referred to as "component (c)") for forming the olefin polymerization catalyst used in the invention can be represented by, for example, the following formula (2):

$$\mathsf{H}^{\mathsf{0}}_{\mathsf{0}}\mathsf{AlX}_{\mathsf{3}\mathsf{0}}$$
 (2)

wherein R⁹ is a hydrocarbon group of 1 to 12 carbon atoms, X is a halogen atom or a hydrogen atom, and n is 1 to 3. In the above formula (2), R⁹ is a hydrocarbon group of 1 to 12 carbon atoms, for example, an alkyl group, a cycloalkyl group or an aryl group. Particular examples thereof include methyl, ethyl, n-propyl, isopropyl, isobutyl, pentyl, hexyl, cyclopentyl, cyclohexyl, phenyl and tolyl.

Examples of the organoaluminum compounds (c) include:

irialkykaluminums, such as trimethykaluminum, triethykaluminum, triisopropykaluminum, triisobutykaluminum, trioc tykaluminum, tri(2-ethylhexyl)aluminum and tridecykaluminum;

alkenylatuminums, such as isoprenylatuminum;

dialkylaluminum halides, such as dimethylaluminum chloride, diethylaluminum chloride, diisopropylaluminum chloride, diisobutylaluminum chloride and dimethylaluminum bromide;

alkylaluminum sesquihalides, such as methylaluminum sesquichoride, ethylaluminum sesquichloride, isopropylaluminum sesquichloride, butylaluminum sesquichloride and ethylaluminum sesquibromide;

alkylaluminum dihalides, such as methylaluminum dichloride, ethylaluminum dichloride, isopropylaluminum dichloride and ethylaluminum dibromide; and

alkylaluminum hydrides, such as diethylaluminum hydride and diisobutylaluminum hydride.

Also employable as the organoaluminum compound (c) is a compound represented by the following formula (3):

$$\mathsf{H}^{9}_{\mathsf{A}}\mathsf{AlL}_{\mathsf{A}\mathsf{B}}$$
 (3)

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The polymerization temperature may be, in the suspension polymerization process, in the range of usually -50 to 100 °C, preferably 0 to 90 °C; in the solution polymerization process, in the range of usually 0 to 250 °C, preferably 20 to 200 °C; and in the gas phase polymerization process, in the range of usually 0 to 120 °C, preferably 20 to 100 °C. The polymerization pressure may be in the range of usually atmospheric pressure to 100 kg/cm², preferably atmospheric pressure to 50 kg/cm². The polymerization reaction can be carried out either batchwise, semicontinuously or continuously. The polymerization can also be carried out in two or more stages under different reaction conditions.

The molecular weight of the resulting propylene-1-butene random copolymer can be regulated by adding hydrogen to the polymerization system or by varying the polymerization temperature or the polymerization pressure.

Polypropylene composition

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The polypropylene composition of the invention comprises:

the polypropylene (A) in an amount of 5 to 95 % by weight, preferably 20 to 90 % by weight, more preferably 40 to 85 % by weight, and

the propylene-1-butene random copolymer (B) in an amount of 95 to 5 % by weight, preferably 80 to 10 % by weight and 60 to 15 % by weight.

The polypropylene composition can be prepared by any conventionally known process for preparing resin compositions, for example, by mell-kneading the polypropylene (A) and the propylene-1-butene random copolymer (B).

The melt flow rate MFR of the polypropylene composition may be in the range of 0.1 to 400 g/10 min, preferably 1 to 100 g/10 min.

The melting point of the polypropylene composition may be in the range of 60 to 165 °C, preferably 80 to 160 °C. In addition to the polypropylene and the propylene-1-butene random copolymer, the propylene composition of the invention may contain additives and other resins so long as the objects of the invention are not marred.

Examples of such additives include nucleating agents, antioxidants, hydrochloric acid absorbers, heat stabilizers, light stabilizers, ultraviolet light absorbers, lubricants, antistatic agents, flame retardants, pigments, dyes, dispersing agents, copper inhibitors, neutralizing agents, toaming agents, plasticizers, antifoaming agents, crosslinking agents, flowability improvers such as peroxides, and weld strength improvers.

As the antioxidants, phenolic antioxidants, sulfur antioxidants, phosphorus antioxidants, etc. are employable.

Examples of the phenolic antioxidants include phenols compounds, such as 2.6-di-ten-butyl-p-cresol, stearyl (3,3-dimethyl-4-hydroxybenzyl)thioglycolate, stearyl-β-(4-hydroxy-3,5-di-ten-butyl-benzyl)propionate, distearyl-3,5-di-ten-butyl-4-hydroxybenzyl phosphonate, 2,4,6-tris(3',5'-di-ten-butyl-4'-hydroxybenzylthio)-1,3,5-triazine, distearyl (4-hydroxy-3-methyl-5-tent-butylbenzyl)malonate, 2,2'-methylenebis(4-methyl-6-tent-butylphenol), 4,4'-methylenebis (2,5-di-tent-butylphenol), 2,2'-methylenebis[6-11-methylcyclohexyl]p-cresol], bis[3,5-bis(4-hydroxy-3-tent-butylphenyl] butyric acid] glycol ester, 4,4'-butylidenebis(6-tent-butyl-m-cresol), 1,1,3-tris(2-methyl-4-hydroxy-5-tent-butylphenyl)butane, bis[2-tent-butyl-4-methyl-6-(2-hydroxy-3-tent-butyl-5-methylbenzyl)phenyl] terephthalate, 1,3,5-tris(2,6-dimethyl-3-hydroxy-4-tent-butyl-4-hydroxy-3-tent-butyl-4-hydroxybenzyl)-2,4,6-trimethylbenzene, tetrakis[methylene-3-(3,5-di-tent-butyl-4-hydroxyphenyl)propionate]methane, 1,3,5-tris(3,5-di-tent-butyl-4-hydroxyphenyl)propionyloxyethyl] isocyanurate, 2-octylthio-4,6-di(4-hydroxy-3,5-di-tent-butyl)phenoxy-1,3,5-triazine and 4,4'-thiobis(6-tent-butyl-m-cresol); and polyhydric phenol carbonic acid oligo esters, such as carbonic acid oligo esters of 4,4'-butylidenebis(2-ten-butyl-5-methylphenot) (e.g., polymerization degree: 2,to_10).

Examples of the suffur antioxidants include dialkyl thiodipropionates, such as dilauryl thiodipropionate, dimyristyl thiodipropionate and distearyl thiodipropionate; and esters of polyhydric alcohols (e.g., glycerol, trimethylolethane, trimethylolpropane, pentaerythritol and trishydroxyethyl isocyanurate) and alkylthiopropionic acids (e.g., butylthiopropionic acid, octylthiopropionic acid, laurylthiopropionic acid and stearylthiopropionic acid), such as pentaerythritotte-tralauryl thiopropionate.

Examples of the phosphorus antioxidants include trioctyl phosphite, trilauryl phosphite, tridecyl phosphite, cctyldiphenyl phosphite, tris(2,4-di-tert-butylphenyl) phosphite, triphenyl phosphite, tris(butoxyethyl) phosphite, tris(nonylphenyl) phosphite, distearylpentaerythritol diphosphite, tetra(tridecyl)-1,1,3-tris(2-methyl-5-tert-butyl-4-hydroxyphenyl)butane diphosphite, tetra(tridecyl)-4,4'-isopropylidenediphenyl diphosphite, tetra(tridecyl)-4,4'-butylidenebis(3-methyl-6-tert-butylphenol) diphosphite, tris(3,5-di-tert-butyl-4-hydroxyphenyl) phosphite, tris(mono-di mixed nonylphenyl) phosphite, hydrogenated 4,4'-isopropylidenediphenol polyphosphite, bis(octylphenyl)-bis(4,4'-butylidenebis (3-methyl-6-tert-butylphenol))-1,6-hexanediol diphosphite, phenyl-4,4'-isopropylidenediphenol-pentaerythritol diphosphite, bis(2,4-di-tert-butylphenyl)pentaerythritol diphosphite, bis(2,4-di-tert-butylphenyl)pentaerythritol diphosphite, tris(4,4'-isopropylidenebis(2-tert-butylphenol)) phosphite, tris(1,3-di-stearoyloxyisopropyl) phosphite, 4,4'-isopropylidenebis(2-tert-butylphenol)-di

(nonylphenyl) phosphite, 9,10-dihydro-9-oxa-1-phosphaphenanthrene-10-oxide and tetrakis(2,4-di-tert-butylphenyl)-4,4'-biphenylene diphosphite.

Examples of further employable antioxidants are 6-hydroxychroman derivatives, such as α -, β -, γ - and δ -tocopherol and mixtures thereof, 2,5-dimethyl-, 2,5,8-trimethyl- and 2,5,7,8-tetramethyl-substituted derivatives of 2-(4-methyl-penta-3-enyl)-6-hydroxychroman, 2,2,7-trimethyl-5-terl-butyl-6-hydroxychroman, 2,2,5-trimethyl-7-terl-butyl-6-hydroxychroman, 2,2,5-trimethyl-6-terl-butyl-6-hydroxychroman, 2,2,5-trimethyl-6-terl-butyl-6-hydroxychroman.

As the hydrochloric acid absorbers, there can be employed mixed salts represented by the formula

MxAy(OH)2x+3y-2z(A)z-8H2O

(wherein M is Mg, Ca or Zn; A is an anion other than hydroxyl group; x, y and z are each a positive number; and a is 0 or a positive number, such as Mg₆Al₂(OH)₁₆CO₃4H₂O.

Mg_kAl₂(OH)₂₀CO₃·5H₂O, Mg₅Al₂(OH)₁₄CO₃·4H₂O, Mg₁₀Al₂(OH)₂₂(CO₃)₂·4H₂O, Mg_kAl₂(OH)₁₆HPO₄·4H₂O, Ca₆Al₂(OH)₁₆CO₃·4H₂O, Zn₆Al₂(OH)₁₆CO₃·4H₂O, Zn₆Al₂(OH)₁₆SO₄·4H₂O, Mg_kAl₂(OH)₁₆SO₃·4H₂O and Mg_kAl₂(OH)₁₆SO₃·3H₂O.

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Examples of the fight stabilizers include hydroxybenzophenones, such as 2-hydroxy-4-methoxybenzophenone, 2-hydroxy-4-n-octoxybenzophenone, 2,2-dihydroxy-4-methoxybenzophenone and 2,4-dihydroxybenzophenone; benzotriazoles, such as 2-(2-hydroxy-3-tert-butyl-5-methylphenyl)-5-chlorobenzotriazole, 2-(2-hydroxy-3-5-di-tert-butylphenyl)-5-chlorobenzotriazole, 2-(2-hydroxy-3-5-di-tert-butylphenyl)-5-chlorobenzotriazole; benzoales, such as phenyl salicylate, p-tert-butylphenyl salicylate, 2,4-di-tert-butylphenyl-3,5-di-tert-butyl-4-hydroxybenzoale and hexadecyl-3,5-di-tert-butyl-4-hydroxybenzoale; nicket compounds, such as 2,2-thiobis (4-tert-octylphenol) Ni salt, [2,2'-thiobis(4-tert-octylphenolate)]-n-butylamine Ni and (3,5-di-tert-butyl-4-hydroxybenzyl)phosphonic acid monoethyl ester Ni salt; substituted acrylonitriles, such as methyl α-cyano-β-methyl-β-(p-methoxyphenyl)acrylate; oxalic acid dianitides, such as N'-2-ethylphenyl-N-ethoxyphenyloxamide and N-2-ethylphenyl-N'-2-ethoxyphenyloxamide; and hindered amine compounds, such as bis(2,2,6,6-tetramethyl-4-piperidine)sebacate, poly [((6-(1,1,3,3-tetramethylbutyl)imino)-1,3,5-triazine-2,4-diyl(4-(2,2,6,6-tetramethylpiperidyl)imino)hexamethylene] and a condensate of 2-(4-hydroxy-2,2,6,6-tetramethyl-1-piperidyl)ethanol and dimethyl succinate.

Examples of the lubricants include aliphatic hydrocarbons, such as parafiin waxes, polyethylene waxes and polypropylene waxes; higher latty acids, such as capric acid, lauric acid, myristic acid, patmitic acid, margaric acid, stearic acid, arachidic acid and behenic acid; metallic salts of these acids, such as lithium salt, calcium salt, sodium salt, magnesium salt and potassium salt; aliphatic alcohols, such as patmityl alcohol, cetyl alcohol and stearyl alcohol; aliphatic amides, such as capronamide, caprylamide, caprinamide, laurylamide, myristylamide, palmitylamide and stearylamide; esters of fatty acids and alcohols; and fluorine compounds, such as fluoroalkylcarboxylic acids or their metallic salts and fluoroalkylsulfonic acid metallic salts.

The above-mentioned additives may be contained in the polypropylene composition in amounts of 0,0001 to 10 % by weight.

By the addition of these additives to the polypropylene composition of the invention, molded articles much more improved in, for example, balance between properties; durability, paintability, printability, mar or scuffing resistance and molding processability can be formed from the composition.

The polypropylene composition of the invention may contain nucleating agents, as described above.

There is no specific limitation on the nucleating agent used herein, and various nucleating agents conventionally known are employable. Of these nucleating agents, preferred are aromatic phosphoric ester salts and dibenzylideneschilds.

As the aromatic phosphoric acid ester salts, there can be used compounds represented by the following formula:

$$\begin{bmatrix}
R^{3} & & & & \\
R^{1} & & & & \\
R^{3} & & & & \\
R^{3} & & & & \\
R^{2} & & & & \\
\end{bmatrix}$$

wherein R¹ is oxygen, sulfur or a hydrocarbon group of 1 to 10 carbon atoms; R² and R³ may be the same or different and are each hydrogen or a hydrocarbon group of 1 to 10 carbon atoms; both R², both R³, or R² and R³ may be bonded together to form a ring; M is a monovalent to trivalent metal atom; and n is an integer of 1 to 3.

Examples of the nucleating agents represented by the above formula include sodium-2,2'-methylene-bis(4.6-di-tbutylphenyl)phosphate, sodium-2,2'-ethylidene-bis(4,6-di-t-butylphenyl)phosphate, lithium-2,2'-methylene-bis(4,6-dit-butylphenyl)phosphate, lithium-2,2'-ethylidene-bis(4,6-di-1-butylphenyl)phosphate, sodium-2,2'-ethylidene-bis(4-ipropyl-6-t-butylphenyl)phosphate, lithium-2,2'-methylene-bis(4-methyl-6-t-butylphenyl)phosphate, lithium-2,2'-methylene-bis(4-ethyl-6-t-butylphenyl)phosphate, calcium-bis[2,2'-thiobis(4-methyl-6-t-butylphenyl)phosphate], calciumbis[2,2'-thiobis(4-ethyl-6-t-butylphenyl)phosphate], calcium-bis[2,2'-thiobis(4,6-di-t-butylphenyl)phosphate], magnesium-bis[2,2'-thiobis(4,6-di-t-butylphenyl)phosphate], magnesium-bis[2,2'-thiobis(4-t-octylphenyl)phosphate], sodium-2,2'-butylidene-bis(4,6-dimethylphenyl)phosphate, sodium-2,2'-butylidene-bis(4,6-di-t-butylphenyl)phosphate, sodium-2,2'-1-octylmethylene-bis(4,6-direthylphenyl)phosphate, sodium-2,2'-1-octylmethylene-bis(4,6-di-1-butylphenyl) calcium-bis[2,2'-methylene-bis(4,6-di-t-butylphenyl)phosphate], magnesium-bis[2,2'-methylene-bis (4,6-di-t-butylphenyl)phosphate), barium-bis[2,2'-methylene-bis(4,6-di-t-butylphenyl)phosphate), sodium-2,2'-methylene-bis(4-methyl-6-t-butylphenyl)phosphate, sodium-2,2'-methylene-bis(4-ethyl-6-t-butylphenyl)phosphate, sodium (4,4'-dimethyl-5,6'-di-t-butyl-2,2'-biphenyl)phosphate. calcium-bis[(4,4'-dimethyl-6,6'-di-t-butyl-2,2'-biphenyl)phosphate], sodium-2,2'-ethylidene-bis(4-m-butyl-6-t-butylphenyl)phosphate, sodium-2,2'-methylene-bis(4,6-dimethylphenyl)phosphate, sodium-2,2'-methylene-bis{4,6-diethylphenyl)phosphate, potassium-2,2'-ethylidene-bis{4,6-di-t-butylphenyl)phosphate, calcium-bis[2,2'-ethylidene-bis(4,6-di-t-butylphenyl)phosphate], magnesium-bis[2,2'-ethylidenebis(4,6-di-t-butylphenyl)phosphate], barium-bis[2,2'-ethylidene-bis(4,6-di-t-butylphenyl)phosphate], aluminum-tris [2,2'-methylene-bis(4,6-di-t-butylphenyl)phosphate] and aluminum-tris[2,2'-ethylidene-bis(4,6-di-t-butylphenyl)phosphate], and mixtures of these compounds. Of these, sodium-2,2-methylene-bis(4,6-di-1-butylphenyl)phosphate is pre-

As the nucleating agents, the following aromatic phosphoric ester salts can also be used:

wherein R⁴ is hydrogen or a hydrocarbon group of 1 to 10 carbon atoms; M is a monovalent to trivalent metal atom; and n is an integer of 1 to 3.

Examples of these compounds include sodium-bis(4-1-butylphenyl)phosphate, sodium-bis(4-methylphenyl)phosphate, sodium-bis(4-ethylphenyl)phosphate, sodium-bis(4-i-propylphenyl)phosphate, sodium-bis(4-t-octylphenyl)phosphate, sodium-bis(4-t-butylphenyl)phosphate, magnesium-bis(4-t-butylphenyl)phosphate, ithium-bis(4-t-butylphenyl)phosphate and aluminum-bis(4-t-butylphenyl)phosphate, and mixtures of these compounds. Of these, sodium-bis(4-t-butylphenyl)phosphate is preferred.

Further, as the nucleating agent, there can also be used benzylidenesolbitols represented by the formula:

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$$R^5 \longrightarrow O \longrightarrow O \longrightarrow O \longrightarrow R^5$$

wherein RS is hydrogen or a hydrocarbon group of 1 to 10 carbon atoms.

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Examples of these compounds include 1,3,2,4-dibenzylidenesorbitol, 1,3-benzylidene-2,4-p-methylbenzylidene-sorbitol, 1,3-benzylidene-2,4-p-ethylbenzylidenesorbitol, 1,3-p-methylbenzylidene-2,4-benzylidenesorbitol, 1,3-p-methylbenzylidene-2,4-benzylidenesorbitol, 1,3-p-methylbenzylidene-2,4-p-ethylbenzylidenesorbitol, 1,3-p-ethylbenzylidene-2,4-p-methylbenzylidene-2,4-p-methylbenzylidene)sorbitol, 1,3,2,4-di(p-ethylbenzylidene)sorbitol, 1,3,2,4-di(p-ethylbenzylidene)sorbitol, 1,3,2,4-di(p-methylbenzylidene)sorbitol, 1,3,2,4-di(p-methylbenzylidene)sorbitol, 1,3,2,4-di(p-methylbenzylidene)sorbitol, 1,3,2,4-di(p-ethylbenzylidene)sorbitol, 1,3,2,4-di(p-ethylbenzylidene)sorbitol, 1,3,2,4-di(p-ethylbenzylidene)sorbitol, 1,3-p-chlorobenzylidene-2,4-p-ethylbenzylidenesorbitol, 1,3-p-chlorobenzylidene-2,4-p-ethylbenzylidenesorbitol, 1,3-p-ethylbenzylidene-2,4-p-ethylbenzylidenesorbitol, 1,3-p-ethylbenzylidene-2,4-p-ethylbenzylidenesorbitol, 1,3-p-ethylbenzylidene-2,4-p-ethylbenzylidenesorbitol, 1,3,2,4-di(p-ethylbenzylidene)sorbitol, 1,3,2,

Further, nucleating agents which may be used are metallic salts of aromatic carboxylic acids and aliphatic carboxylic acids. Examples thereof include aluminum benzoate, aluminum p-t-butylbenzoate, sodium adipate, sodium thiophenecarboxylate and sodium pytrolecarboxylate.

inorganic compounds such as talc are also employable as the nucleating agents.

The nucleating agent as described above may be contained in the polypropylene composition of the invention in an amount of 0.001 to 10 % by weight, preferably 0.01 to 5 % by weight, more preferably 0.1 to 3 % by weight.

By the addition of the nucleating agent, the crystallization rate of the polypropylene composition can be improved and the crystal particles can be made smaller in the crystallization process, whereby high-speed molding can be attained.

As the other resins which may be added to the composition of the invention, thermoplastic resins or thermosetting resins are employable. Examples of such resins include α-olefin homopolymers such as polyethylene and poly1-butene, α-olefin copolymers, copolymers of α-olefins and vinyl monomers, modified olefin polymers such as maleic anhydride-modified polypropylenes, polyamides, polycarbonates, ABS resins, polystyrenes, polyvinyl chlorides, polyphenylene oxides, petroleum resins and phenolic resins.

The polypropylene composition of the invention may further contain inorganic fillers. Examples of the inorganic fillers include:

powdery fillers, for example, natural silicic acids or silicates such as fine-powdery talc, kaolinite, calcined clay, pyrophyllite, sericite and wollastonite, carbonates such as precipitated calcium carbonate, ground limestone and magnesium carbonate, hydroxides such as aluminum hydroxide and magnesium hydroxide, oxides such as zinc oxide, zinc white and magnesium oxide, and synthetic silicic acids or silicates such as hydrated calcium silicate, hydrated aluminum silicate, hydrated silicic acid and silicic anhydride;

flaky filters, such as mica;

fibrous fillers. such as basic magnesium sulfate whisker, calcium titanate whisker, aluminum borate whisker, sepiolite, PMF (Processed Mineral Fiber), xonotolite, potassium titanate and ellestadile; and balloon fillers, such as glass balloon and fly ash balloon.

Ot these filters, fine-powdery talc is preferably used in the invention. Particularly preferred is fine-powdery talc having a mean particle diameter of 0.2 to 3 μ m, especially 0.2 to 2.5 μ m.

The teld desirably contains particles having a mean diameter of not less than 5 µm in an amount of not more than 10 % by weight, preferably not more than 8 % by weight.

The mean particle diameter of the talc can be measured by a liquid phase precipitation method.

In the present invention, it is preferable that the talc has a mean aspect ratio (ratio of longitudinal or lateral length to thickness) of not less than 3, particularly not less than 4.

The inorganic fillers, in particular tate may be used with or without surface treatment in the invention. The surface treatment can be carried out chemically and physically, for example, using treating agents such as silane coupling

agents, higher fatty acids, fatty acid metallic salts, unsaturated organic acids, organic titanates, resin acids and polyethylene glycols.

By the use of the surface-treated inorganic fillers such as talc, a polypropylene composition excellent also in weld strength, paintability and molding processability can be obtained.

The inorganic filters as mentioned above may be used in combination.

If necessary, organic fillers such as high styrenes, lignin and reclaimed rubbers may be used in combination with the inorganic fillers.

The polypropylene composition of the invention described above is excellent in heat resistance and low-temperature heat-sealing properties as well as in flexibility and impact resistance. The polypropylene composition can be tavorably used for heat-sealable OPP, injection molded articles, e.g., container, improved in impact resistance, stretch-blow molded articles, containers for retort pouch toods, films and sheets.

Further, the polypropylene composition of the invention can be applied to other various uses than the abovementioned ones. For example, it can be used for household appliance parts such as housings and tubs of washing machines, automobile interior parts such as trims, interior panels and column covers, automobile exterior parts such as fenders, bumpers, side motes, mudguards and mirror covers, and ordinary miscellaneous goods.

Next, the polypropylene composite film of the invention comprising a layer of the above-mentioned polypropylene composition is described in detail.

Polypropylene composite film

The polypropylene composite film according to the invention comprises:

(I) a crystalline polypropylene layer, and

(II) a layer formed from the polypropylene composition of the present invention, which is laminated on at least one surface of the crystalline polypropylene layer (I).

(I) Crystalline polypropylene layer

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The substrate of the polypropylene composite film of the invention is formed from a crystalline polypropylene (I). In the present invention, as the crystalline polypropylene, polypropylenes conventionally known as film-forming materials can be used, but preferably used is a polypropylene having an isotactic index I.I. (boiling n-heptane-insoluble component) of not less than 75 %, preferably 75 to 99 %.

The crystalline polypropylene preferably has a density of 0.89 to 0.92 g/cm³ and a melt index (230 °C) of 0.1 to 10 g/10 min.

Although a homopolypropylene is generally used as the crystalline polypropylene, a propylene random copolymer containing units from an olefin other than propylene in a small amount, e.g., not more than 5 % by mol, may be used as long as the objects of the invention are not marred. Examples of the other olefins include α-olefins of 2 to 20 carbon atoms except propylene, such as ethylene, 1-butene, 1-pentene, 1-hexane, 1-heptene, 1-octene, 1-decene, 1-decene, 1-hexandecene and 4-methyl-1-pentene.

The crystalline polypropylene used in the invention can be prepared by a conventional process using a known solid titanium catalyst component or metallocene compound catalyst component.

To the crystalline polypropylene, additives such as heat stabilizers, ultraviolet light absorbers, anti-blocking agents, slip agents and antioxidants may be added.

In the invention, the crystalline polypropylene layer (I) may be either unstretched or biaxially stretched.

(II) Polypropylene composition layer

The heat-sealing layer of the polypropylene composite film is formed from the polypropylene composition of the invention described above.

Among the polypropylene compositions, particularly preferably used to form the heat-sealing layer of the polypropylene composite film is a composition comprising (A) a polypropylene and (B) a propylene-1-butene random copolymer, both described below.

The polypropylene (A) is preferably a crystalline polypropylene containing units from propylene in an amount of not less than 90 % by mol.

The propylene-1-butene random copolymer (B) desirably have the following properties:

(1) The copolymer contains units from propylene in an amount of 50 to 95 % by mol, preferably 55 to 93 % by mol, more preferably 60 to 90 % by mol, and contains units from 1-butene in an amount of 5 to 50 % by mol, preferably

7 to 45 % by mol, more preferably 10 to 40 % by mol.

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- (2) The intrinsic viscosity (η) is in the range of 0.1 to 5 df/g, preferably 0.5 to 4 df/g, more preferably 1 to 3 df/g.
- (3) The molecular weight distribution (Mn/Mw) is not more than 3, preferably not more than 2.5.
- (4) The randomness parameter B value is in the range of 1.0 to 1.5, preferably 1.0 to 1.3, more preferably 1.0 to 1.2.
- (5) The meiting point Tm, as measured by a differential scanning calorimeter, is in the range of 60 to 140°C, preferably 80 to 120°C, more preferably 90 to 110°C.
- (6) The crystallinity, as measured by X-ray diffractometry, is preferably in the range of 15 to 65 %, more preferably 20 to 60 %.

Further, the propylene-1-butene random copolymer may contain regio-irregular units based on 2,1-insertions of propylene in an amount of not less than 0.05 %.

The polypropylene composition for forming the heat-sealing layer of the polypropylene composite film of the invention contains the propylene-1-butene random copolymer (B) having the above properties, so that it exhibits excellent heat-sealing properties. For example, it the intrinsic viscosity (η) , i.e., the above property (2) of the propylene-1-butene copolymer, exceeds 5 dVg, the moldability of the composition may be reduced and thus it may be difficult to form a heat-sealing layer having a desired thickness, e.g., not more than 50 μ m. If the intrinsic viscosity (η) is less than 0.1 dVg, the heat-sealing strength may be lowered.

If the melting point, i.e., the above property (5), exceeds 140 °C, heat-sealing may require high temperatures, e. g., not lower than 130 °C. If the melting point is lower than 60 °C, the low-temperature heat-sealing properties are improved, but the scratch resistance may be lowered and blocking of the resulting films may take place during storage, making their practical use hard.

A propylene-1-butene random copolymer having a crystallinity of 15 to 65 % is preferred because a film of excellent low-temperature heat-sealing properties and high blocking resistance can be obtained. If the crystallinity is less than 15 %, the resulting film may have insufficient scratch resistance and tends to suffer blocking and tackiness. If the crystallinity is more than 65 %, the low-temperature heat-sealing properties are not sufficiently improved.

In the present invention, the propylene-1-butene random copolymer which is partially or fully modified with unsaturated carboxylic acids or anhydrides thereof can also be used. The modified product of the propylene-1-butene random copolymer is excellent in overwrapping properties and bond properties to other materials such as metals.

In the polypropylene composite film of the invention, the heat-sealing layer (II) is formed from a polypropylene composition comprising the polypropylene (A) and the propylene-1-butene random copolymer (B), and the amounts of these components (A) and (B) can be properly determined depending on the purposes.

In the invention, for example, the heat-sealing layer (II) of the polypropylene composite film may be formed from a polypropylene composition which contains the polypropylene (A) in an amount of 5 to 50 % by weight and the propylene-1-butene random copolymer in an amount of 95 to 50 % by weight.

From the polypropylene composition containing the components (A) and (B) in the above-mentioned amounts, there can be obtained a polypropylene composite film (1) excellent in not only transparency, low-temperature heat-sealing properties and hot tack properties but also stip properties and mechanical strength such as blocking resistance and scratch resistance.

The heat-sealing layer (II) of the polypropylene composite film of the invention may also be formed from a polypropylene composition which contains the polypropylene (A) in an amount of not less than 50 % by weight and less than 90 % by weight and the propylene-1-butene random copolymer (B) in an amount of not less than 10 % by weight and less than 50 % by weight.

From the polypropylene composition containing the components (A) and (B) in the above-mentioned amounts, there can be obtained a polypropylene composition film (2) excellent in not only transparency and low-temperature heat-sealing properties but also mechanical strength such as blocking resistance and scratch resistance.

In the polypropylene composite films according to the invention, if the amount of the crystalline polypropylene (A) in the polypropylene composition exceeds 90 % by weight, the softening temperature of the film strikingly rises and the low-temperature heat-sealing properties are markedly lowered. If the amount thereof is less than 5 % by weight, the blocking resistance and the scratch resistance of the film may be lowered.

The polypropylene composite film according to the invention comprises the crystalline polypropylene layer (I) and the polypropylene composition heat-sealing layer (II), and the heat-sealing layer (II) may be provided on one or both surfaces of the crystalline polypropylene layer (I) serving as a substrate.

In the polypropylene composite film of the invention, the crystalline polypropylene layer (I), as a substrate may be either unstretched, monoaxially or biaxially stretched. One or both surfaces of the substrate (I) may be subjected to corona discharge treatment by a conventional method.

The composite film of the invention can be obtained from the crystalline polypropylene (I) and the polypropylene composition (II) by, for example, the following processes.

(1) The crystalline polypropylene (I) and the polypropylene composition (II) are co-extruded to form a laminated sheet.

This laminated sheet may be monoaxially stretched in the longitudinal or lateral direction, or may be biaxially stretched in the longitudinal and lateral directions separately or simultaneously.

(2) The crystalline polypropylene (I) is melt-extruded to form a film (substrate), on which the polypropylene composition (II) is melt-extruded to form a laminated sheet.

This laminated sheet may be monoaxially stretched in the longitudinal or lateral direction, or may be biaxially stretched in the longitudinal and lateral directions separately or simultaneously.

- (3) The crystalline polypropytene (I) is melt-extruded and monoaxially stretched in the longitudinal or lateral direction to form a substrate. On the monoaxially stretched substrate, the polypropytene composition (II) is melt-extruded, or a film of the polypropytene composition (II) preliminarily formed is laminated. The resulting laminated sheet is stretched in the unstretched direction of the substrate.
- (4) The crystalline polypropylene (I) is melt-extruded and biaxially stretched in the longitudinal and lateral directions separately or simultaneously to form a substrate. On the biaxially stretched substrate, the polypropylene composition (II) is melt-extruded, or a film of the polypropylene composition (II) preliminarily formed is laminated.
- (5) A film of the crystalline polypropylene (I) and a film of the polypropylene composition (II) are laminated together with an adhesive. The resulting taminated sheet may be monoaxially stretched in the longitudinal or lateral direction, or may be biaxially stretched in the longitudinal and lateral directions separately or simultaneously.

In the polypropylene composite film according to the invention, the layer of crystalline polypropylene (I) has a thickness of 5 to 200 µm, preferably 10 to 60 µm, and the layer of polypropylene composition (II) has a thickness of 0.1 to 50 µm, preferably 0.5 to 20 µm.

In the composite film of the invention, when the substrate made of the crystalline polypropylene (I) is stretched, the stretch ratio in the longitudinal direction is in the range of 3 to 7 times, preferably 4 to 6 times, and the stretch ratio in the lateral direction is in the range of 3 to 12 times, preferably 6 to 10 times.

EFFECT OF THE INVENTION

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The polypropylene composition according to the invention is excellent in heat resistance and low-temperature heat-sealing properties as well as in flexibility and impact resistance. The polypropylene composition can be favorably used for heat-sealable OPP, injection molded articles such as containers improved in impact resistance, stretch-blow molded articles, containers for retort pouch foods, films and sheets.

Further, the polypropylene composition of the invention can be applied to other various uses than the abovementioned ones. For example, it can be used for household appliance parts such as housings and tubs of washing machines, automobile interior parts such as trims, interior panels and column covers, automobile exterior parts such as fenders, bumpers, side moles, mudguards and mirror covers, and ordinary miscellaneous goods.

The polypropylene composite film of the invention having a heat-sealing layer of the polypropylene composition is excellent in heat-sealing properties. Particularly, the film can be heat-sealed in a wide temperature range, in particular at low temperatures, and has high heat-sealing strength. Further, even when the polypropylene composite film is stored for a long period of time, its heat-sealing temperature does not change with time, and thus stable heat-sealing operation can be assured.

Furthermore, because of its excellent transparency, scratch resistance and blocking resistance, the polypropylene composite film of the invention can be used for high-speed packaging.

According to the invention, a polypropylene composite film further having good hot tack properties in addition to the above-mentioned excellent properties can be formed.

The polypropylene composite film according to the invention can be favorably used for packaging of foods, fill-packaging and fiber packaging.

EXAMPLE

The present invention will be further described with reference to the following examples, but it should be construed that the invention is in no way limited to those examples.

Measurement of properties of the propylene-1-butene random copolymer (B)

(1) Propylene content and 1-Butene content:

Determined using 13C-NMR.

(2) Intrinsic viscosity (η):

Measured in decahydronaphthalene at 135 °C and expressed by dVg.

(3) Molecular weight distribution (Mw/Mn):

Measured in the following manner using GPC-150C produced by Millipore Co.

A separatory column of TSK-GNH-HT having a diameter of 27 mm and a length of 600 mm was used, and the column temperature was set to 140 °C. A mobile phase used was o-dichlorobenzene (available from Wako Junyaku Kogyo K.K.) containing 0.025 % by weight of BHT (available from Takeda Chemical Industries, Ltd.) as an antioxidant and a moving rate was 1.0 ml/min. 500 µl of a solution containing 0.1 % by weight of a sample was introduced into the column. A differential refractometer was used as a detector.

With regard to molecular weight standards, polystyrenes available from Toso Co., Ltd. were used for Mw < 1,000 and Mw > 4×10^6 , and polystyrenes available from Pressure Chemical Co. were used for 1,000 < Mw < 4×10^6 .

(4) B value (monomer distribution)

A homogeneous sample solution of about 200 mg of a copolymer in 1 mf of hexachlorobutadiene was measured for its 13 C NMR spectrucm in a sample tube having a diameter of 10 mm under the measuring conditions of usually a temperature of 120 °C, a frequency of 25.05 MHz, a spectral width of 1,500 Hz, a fifter width of 1,500 Hz, pulse repetition time of 4.2 sec and integrating times of 2,000 to 5,000. From the spectrum obtained, P_E , P_O and P_{OE} were sought, and the B value was calculated using P_E , P_O and P_{OE} .

(5) Metting point (Tm):

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Measured in the following manner using a DSC-7 type measuring device produced by Perkin Elmer Co. About 5 mg of a sample charged in an aluminum pan was heated to 200 °C, kept at 200 °C for 5 minutes, then cooled to -40 °C at a rate of 10 °C/min and heated again at a rate of 10 °C/min, to obtain an endothermic curve of the sample. From the endothermic curve, the melting point (Tm) was determined.

(6) Crystallinity:

Determined by X-ray diffractometry for a press sheet having a thickness of 1.0 mm which was stored for at least 24 hours after the press molding.

(7) Triad tacticity

A 13C-NMR spectrum (tetramethylsilane standard) of a hexachlorobutadiene solution of a sample was measured, and a proportion (%) of the area of peaks appeared in the region of 21,0 to 21.0 ppm to the total area (100 %) of peaks appeared in the region of 19.5 to 21.9 ppm was calculated.

(8) Proportion of hetero bonds based on 2,1-insertions:

— Determined in the aloresaid manner using a ¹³C-NMR spectrum with reference to "Polymer", <u>30</u>, 1350 (1989).

Polypropylenes and propylene-1-butene random copolymers used in Examples 1 to 6 and Comparative Examples 1 and 2 are described below.

In the examples and comparative examples, the following polypropylenes prepared using a general-purpose solid titanium calalyst component were used.

Polypropylene-1: a propylene homopolymer (intrinsic viscosity (η): 2.9 d/g, DSC melting point: 164 °C, crystallinity: 62 %).

Polypropylene-2: a propylene random copolymer (composition: propylene = 96.4 % by mol, ethylene = 2.1 % by mol, 1-butene = 1.5 % by mol, intrinsic viscosity (η): 2.1 d/g, DSC melting point: 142 °C, crystallinity: 56 %).

Next, a synthesis example for synthesizing a catalyst component, and preparation examples for preparing the propylene-1-butene random copolymers (B) using a catalyst component synthesized by the same procedure as in the synthesis example are described. The properties of the propylene-1-butene random copolymers (PBR) obtained in the preparation examples are set forth in Table 2.

Synthesis Example 1

Synthesis of rac-dimethylsilylene-bis[1-(2-ethyl-4-phenylindenyl)]zirconium dichloride

Synthesis of 3-(2-biphenylyl)-2-ethylpropionic acid:

To a 500-ml four-necked round flask (equipped with a stirrer, a Dimroth condenser, a dropping lunnel and a thermometer) were introduced 13.46 g (120 mmol) of potassium t-butoxide, 100 ml of toluene and 20 ml of N-methylpyr-rolidone, and then, in a nitrogen atmosphere, a solution of 20.7 g (110 mmol) of diethyl ethylmalonate in 50 ml of toluene was dropwise added with heating to 60 °C. After the dropwise addition was completed, the mixture was reacted for 1 hour at the same temperature, and then, at the same temperature, a solution of 20.27 g (100 mmol) of 2-phenylbenzyl bromide in 30 ml of toluene was dropwise added. After the dropwise addition was completed, the mixture was heated and refluxed for 2 hours. The reaction mixture was poured into 200 ml of water, and 2N-HCl was added to adjust pH of the mixture to 1. The organic phase was separated, and the aqueous phase was extracted three times with 100 ml of toluene. The combined organic phases were washed with a saturated saline solution until the organic phases became neutral, tollowed by drying over anhydrous Na₂SO₄. The solvent was removed under reduced pressure to obtain 36.7 g of an yellow orange liquid concentrate.

To a 1-liter four-necked round flask (equipped with a stirrer, a Dimroth condenser, a dropping funnel and a thermometer) were introduced 67.3 (1.02 mol) of potassium hydroxide and 150 ml of a methanol aqueous solution (methanol/water = 4/1 (v/v)), and then, in a nitrogen atmosphere at room temperature, a solution of the concentrate obtained above in 50 ml of a methanol aqueous solution (methanol/water = 4/1 (v/v)) was dropwise added. After the dropwise addition was completed, the mixture was heated and refluxed for 4 hours, and then cooled to room temperature, and the solid precipitated was filtered. The solid was dissolved in water, and sulfuric acid was added to the solution until it became acidic (pH = 1). The solution was extracted five times with 100 ml of methylene chloride. The combined organic phases were dried over anhydrous Na₂SO₄, and then, the solvent was removed under roduced pressure to obtain 24.2 g of a white solid product.

Subsequently, to a 300-ml three-necked round flask (equipped with a stirrer tip, a Dimroth condenser and a thermometer) were introduced 24.2 g of the white solid obtained above, 56 ml of acetic acid, 37 ml of water and 13.1 ml of concentrated sulfuric acid, and they were refluxed for 6 hours in a nitrogen atmosphere. After the reaction was completed, the acetic acid was distilled off under reduced pressure, 50 ml of water was added, and the mixture was extracted three times with 50 ml of methylene chloride. The combined organic phases were washed with 50 ml of a saturated saline solution and dried over anhydrous Na_2SO_4 . Then, the solvent was distilled off under reduced pressure, and the residue was purified by silica gel chromatography using a gradient of hexane/ethyl acetate (2/1 \rightarrow 1/1, by volume) to obtain 13.7 g of a white solid (yield: 54 %). The properties of the product thus obtained are described below

```
FD-MS: 254 (M*)
mp.: 91.2 - 94.0 °C
NMR (CDCl<sub>3</sub>- 90 Hz):
δ = 0.71 (t, J = 7.2 Hz, 3H, CH<sub>3</sub>);

1.16 - 1.58 (m, 2H);
2.32 (bquin, J = 7.0 Hz, 1H, >CH-);
2.61 - 2.99 (m, 2H);
6.89 - 7.47 (m, 9H).

IR (KBr disk): 1696 cm<sup>-1</sup> (v<sub>c=0</sub>).
```

Synthesis of 3-(2-biphenylyl)-2-ethylpropionyl chloride:

To a 100-ml three-necked round flask (equipped with a stirrer tip, a Dimroth condenser, a thermometer and a trap of NaOH) were introduced 13.3 g (52.4 mmol) of 3-(2-biphenylyl)-2-ethylpropionic acid and 25.9 ml (355 mmol) of thionyl chloride, and they were refluxed for 2.5 hours with heating in a nitrogen atmosphere. After the reaction was completed, the unreacted thionyl chloride was distilled off under reduced pressure to obtain 15.2 g of a crude product as a yellow orange liquid. This acid chloride was used for the next reaction without further purification. The properties of the product thus obtained are described below.

IR (Neal): 1786 cm-1 (vc-v).

45

Synthesis of 4-ethyl-2-phenyl-1-indanone:

To a 200-ml three-necked round flask (equipped with a stirrer tip, a Dimroth condenser, a dropping lunnel, a thermometer and a trap of NaOH) were introduced 8.04 g (60.3 mmol) of anhydrous aluminum chloride and 50 ml of carbon disultide, and then, in a nitrogen atmosphere under ice cooling, a solution of 15.2 g (52.4 mmol) of 3-(2-biphenylyl)-2-ethylpropionyl chloride obtained above in 21 ml of carbon disultide was dropwise added. After the dropwise addition was completed, the internal temperature was elevated to room temperature to perform reaction for 1 hour. Then, the reaction solution was poured into 200 ml of ice water to decompose the unreacted aluminum chloride and extracted twice with 100 ml of ether. The combined organic phases were washed with 100 ml of a saturated NaHCO₃ equeous solution and then with 100 ml of a saturated saline solution, followed by drying over anhydrous Na₂SO₄. The solvent was distilled off under reduced pressure, and the residue was purified by silica gel chromatography using hexane/ethyl acetate (10/1, by volume) to obtain 10.8 g of the aimed product as a yellow solid (yield: 88 %). The properties of the product thus obtained are described below.

```
15 NMR (CDCl<sub>3</sub>, 90 Hz):

δ = 0.98 (1, J = 7.2 Hz, 3H, CH<sub>3</sub>);

1.60 - 2.20 (m, 2H);

2.42 - 2.82 (m, 1H, >CH-);

2.80 (dd, J = 3.8 Hz, 16.5 Hz, 1H);

3.36 (dd, J = 7.6 Hz, 16.5 Hz, 1H);

7.09 - 7.91 (m, 8H).

IR (Neat): 1705 cm<sup>-1</sup> (ν<sub>cm</sub>).
```

Synthesis of 2-ethyl-1-hydroxy-4-phenylindan:

To a 200-ml three-necked round flask (equipped with a stirrer tip, a Dirnroth condenser, a dropping funnel and a thermometer) were introduced 0.85 g (22.6 mmol) of sodium boron hydride and 28 ml of ethanol, and then in a nitrogen atmosphere at room temperature, a solution of 10.6 g (45.1 mmol) of 2-ethyl-4-phenyl-1-indanone in 20 ml of ethanol was dropwise added. After the dropwise addition was completed, the temperature was elevated to 50 °C to perform reaction for further 3.5 hours. After the reaction, the mixture was cooled, and acetone was dropwise added to decompose the unreacted sodium boron hydride. Then, the reaction mixture was concentrated under reduced pressure and extracted with 50 ml of water and 50 ml of ether. The organic phase was separated, and the aqueous phase was extracted twice with 50 ml of ether. The combined organic phases were washed with 100 ml of a saturated saline solution and dried over anhydrous Na₂SO₄. Then, the solvent was distilled off under reduced pressure to obtain 10.67 g of the aimed product (a mixture of two isomers) as a viscous, light yellow liquid (yiekd: 99 %). The properties of the product thus obtained are described below.

```
NMR (CDCl<sub>3</sub>, 90 Hz):

δ = 1.02 (I, J = 7.1 Hz, 3H, CH<sub>3</sub>);

1.31 - 3.28 (m, 2H);

4.86, 5.03 (each d, each J = 6.4 Hz, J = 5:1 Hz, together 1H, >CH-O-);

7.10 - 7.66 (m, 8H).

IR (Neat): 3340 cm<sup>-1</sup> (ν<sub>OH</sub>).
```

Synthesis of 2-ethyl-4-phenylindene:

To a 300-ml four-necked round flask (equipped with a stirrer tip, a dropping lunnel and a thermometer) were introduced 9.78 g (41.3 mmol) of 2-ethyl-1-hydroxy-4-phenylindan, 17.2 ml (123.8 mmol) of triethylamine, 0.25 g (2.1 mmol) of 4-dimethylaminopyridine and 98 ml of methylene chloride, and then in a nitrogen atmosphere under ice cooling, a solution of 6.4 ml (82.5 mmol) of methanesultonyl chloride in 6.5 ml of methylene chloride was dropwise added slowly. After the dropwise addition was completed, the reaction was performed for further 3.5 hours at the same temperature. The reaction mixture was poured into 250 ml of ice water, the organic phase was separated, and the aqueous phase was extracted twice with 50 ml of methylene chloride. The combined organic phases were washed with a saturated NaHCO₃ aqueous solution and then with a saturated saline solution, followed by drying over anhydrous Na₂SO₄. The

solvent was distilled off under reduced pressure, and the residue was purified by silica gel chromatography using hexane to obtain 6.56 g of the aimed product (a mixture of isomers) as a light yellow liquid (yield: 73 %). The properties of the product thus obtained are described below.

```
    5 NMR (CDCl<sub>3</sub>, 90 MHz):
    δ = 1.20 (I, J = 7.6 Hz, 3H, CH<sub>3</sub>);
    2.49 (q, J = 7.6 Hz, 2H);
    3.41 (s. 2H);
    6.61, 6.72 (each bs, together 1H);
    7.09 - 8.01 (m, 8H).
```

Synthesis of dimethylsilylene-bis(2-ethyl-4-phenylindene:

To a 200-ml three-necked round flask (equipped with a stirrer tip, a Dimroth condenser, a dropping funnel and a thermometer) were introduced 5.0 g (22.8 mmol) of 2-ethyl-4-phenylindene, 80 mg (0.63 mmol) of copper thiocyanate and 50 ml of anhydrous ether, and then in a nitrogen atmosphere under ice cooling, a solution of 15.7 ml (25.1 mmol) of n-butyllithium in hexane (concentration: 1.6 M) was dropwise added slowly. After the dropwise addition was completed, the temperature of the mixture was elevated to room temperature to perform reaction for further 1 hour. Then, a solution of 1.52 ml (12.6 mmol) of dimethyldichlorosilane in 4.5 ml of anhydrous ether was dropwise added slowly. After the dropwise addition was completed, the reaction was performed for further 12 hours at room temperature. The reaction mixture was filtered through Celite, and the filtrate was poured into 50 ml of a saturated ammonium chloride aqueous solution. The organic phase was separated, and the equeous phase was extracted with 50 ml of ether. The combined organic phases were washed with a saturated saline solution and dried over anhydrous Na₂SO₄. The solvent was distilled off under reduced pressure, and the residue was separated by silica gel chromatography using a gradient of hexane/methylene chloride (20/1, by volume) to obtain 4.5 g of the aimed product (a mixture of two isomers) as a light yellow solid (yield: 80 %). The properties of the product thus obtained are described below.

```
NMR (CDCl<sub>3</sub>, 90 Hz);
δ = -0.23, -0.17 (each s, together 6H, Si-CH<sub>3</sub>);

1.12, 1.19 (each t, each J = 7.4 Hz, together 6H, CH<sub>3</sub>);
2.44 (bq, J = 7.4 Hz, 4H);
3.81 (s, 2H, >CH-Si-);
6.75 (bs, 2H, 3-H-Ind);
6.88 - 7.74 (m, 16H).
```

Synthesis of rac-dimethylsitylene-bis[1-(2-ethyl-4-phenylindenyl)]zirconium dichloride:

To a 50-ml three-necked round flask (equipped with a stirrer tip, a bead condenser, a dropping tunnel and a thermometer) were introduced 0.84 g (1.69 mmol) of dimethylsilylene-bis(2-ethyl-4-phenylindene) and 17 ml of anhydrous ether, and then at room temperature, a solution of 2.25 ml (3.56 mmol) of n-butyllithium in hexane (concentration: 1.58 M) was dropwise added slowly. After the dropwise addition was completed, the reaction was performed for further 13.5 hours, the reaction solution was cooled to -70 °C in a dry ice/acetone bath, 0.395 g (1.69 mmol) of a ZrCl_k powder was added in pontions, and the resulting mixture was stirred overnight. Then, the solvent was distilled off at room temperature under reduced pressure, 30 ml of methylene chloride was added, and the insoluble matters were filtered off. The filtrate was concentrated at room temperature to precipitate a solid which was filtered, washed twice with 3 ml of anhydrous ether and dried under reduced pressure to obtain 0.17 g of the aimed product as an orange yellow solid (yield: 15 %). The properties of the product thus obtained are described below.

```
NMR (CDCl<sub>3</sub>, 90 MHz);

δ = 1.09 (I, J = 7.3 Hz, 6H, CH<sub>3</sub>);

1.34 (s, 6H, Si-CH<sub>3</sub>);

2.46 (quin, J = 7.3 Hz, 2H);

2.73 (quin, J = 7.3 Hz, 2H);

6.96 (s, 2H, 3-H-Ind);

6.99 - 7.88 (m, 16H).
```

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Preparation Example 1

Preparation of PBR-1

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To a 2-liter autoclave thoroughly purged with nitrogen were introduced 900 ml of hexane and 60 g of 1-butene, followed by adding 1 mmol of triisobutylatuminum. After the temperature of the system was elevated to 70 °C, propylene was fed so that the total pressure became 7 kg/cm²-G, and 0.30 mmol of methylatuminoxane and 0.001 mmol (in terms of 2r atom) of rac-dimethylsitylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride prepared in a manner similar to the above synthesis example were further added. The polymerization was performed for 30 minutes while propylene was continuously fed to keep the total pressure at 7 kg/cm²-G. After the polymerization, the autoclave was released, and a polymer was recovered in a targe amount of methanol and dried in vacuum at 110 °C for 12 hours.

The amount of the polymer (propylene-1-butene random copolymer) obtained was 39.7 g, and the polymerization activity was 79 kg-polymer/mmol-Zr-hr. This polymer had a 1-butene content of 26.4 % by mol, an intrinsic viscosity (η) of 1.60 dl/g and a melting point of 88.4 °C. The proportion of the hetero bonds based on the 2,1-insertions was about 0.02 %. Other properties of the polymer are set forth in Table 2.

Preparation Example 2

Preparation of PBR-2

To a 2-liter autoclave thoroughly purged with nitrogen were introduced 900 ml of hexane and 60 g of 1-butene, followed by adding 1 mmol of trisobutylakuminum. After the temperature of the system was elevated to 70 °C, propylene was fed so that the total pressure became 7 kg/cm²-G, and 0.30 mmol of methylatuminoxane and 0.001 mmol (in terms of Zr atom) of rac-dimethylsitylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride prepared in a manner similar to the above synthesis example were further added. The polymerization was performed for 30 minutes white propylone was continuously fed to keep the total pressure at 7 kg/cm²-G. After the polymerization, the autoclave was released, and a polymer was recovered in a large amount of methanol and dried in vacuum at 110 °C for 12 hours.

The amount of the polymer (propylene-1-butene random copolymer) obtained was 45.2 g, and the polymerization activity was 90 kg-polymer/mmol-Zr-hr. This polymer had a 1-butene content of 20.2 % by mol, an intrinsic viscosity (η) of 1.90 dl/g and a melting point of 101.5 °C. The proportion of the hetero bonds based on the 2,1-insertions was about 0.02 %.

Preparation Example 3

95 Preparation of PBR-3

To a 2-liter autoclave thoroughly purged with nitrogen were introduced 950 ml of hexane and 30 g of 1-butene, followed by adding 1 mmol of triisobutylaluminum. After the temperature of the system was elevated to 70 °C, propylene was fed so that the total pressure became 7 kg/cm²-G, and 0.30 mmol of methylaluminoxane and 0.001 mmol (in terms of 2r atom) of rac-dimethylsitylene-bis{1-(2-methyl-4-phenylindenyl)}zirconium dichloride prepared in a manner similar to the above synthesis example were further added. The polymerization was performed for 30 minutes white propylene was continuously fed to keep the total pressure at 7 kg/cm²-G. After the polymerization, the autoclave was released, and a polymer was recovered in a targe amount of methanol and dried in vacuum at 110 °C for 12 hours.

The amount of the polymer (propylene-1-butene random copolymer) obtained was 52.1 g, and the polymerization activity was 104 kg-polymer/mmol-2i-hi. This polymer had a 1-butene content of 13.9 % by mol, an intrinsic viscosity (η) of 2.51 d/g and a melting point of 116.3 °C. The proportion of the hetero bonds based on the 2,1-insertions was about 0.02 %.

Comparative Preparation Example 1

Preparation of PBR-4

To a 2-liter autoclave thoroughly purged with nitrogen were introduced 830 ml of hexano and 100 g of 1-buteno, tollowed by adding 1 mmol of triisobutylaluminum. After the temperature of the system was elevated to 70 °C, propylene was fed so that the total pressure became 7 kg/cm²-G, and 1 mmol of triethylaluminum and 0.005 mmol (in terms of Ti atom) of a titanium catalyst supported on magnesium chloride was added. The polymerization was performed for 30 minutes while propylene was continuously fed to keep the total pressure at 7 kg/cm²-G. After the polymerization, the autoclave was released, and a polymer was recovered in a large amount of methanol and dried in vacuum at 110

(4) Blocking resistance

Evaluated in accordance with ASTM D1893. A film sample for measuring the heal-sealing strength (1) was cut to give a specimen having a width of 10 cm and a length of 15 cm. Two of the specimens were superposed in such a manner that the polypropylene composition sides faced each other, and they were interposed between two glass plates. The specimens with the glass plates were allowed to stand in an air oven at 50 °C with applying a load of 10 kg. After one day, the specimens were taken out, and the peel strength was measured by a universal testing machine. The peel strength per 1 cm was taken as a value of the blocking resistance.

10 Comparative Examples 1 - 2

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A laminated film was obtained in the same manner as in Example 1 except that the polypropylene composition shown in Table 3 was used. The results are set forth in Table 3.

	T:

	Composition (weight ratio)	Lowest effective heat-seatable temp. (°C)	Haze (%)	Slip properties *	Blocking resistance (g/cm)
Ex. 1	PBR-1/PP-2 (75/25)	110	0.8	0.28/0.28	0.1
Ex. 2	PBR-2/PP-2 (75/25)	100	0.6	0.32/0.31	0.2
Ex. 3	PBR-3/PP-2 (75/25)	88	1.4	0.99/0.51	0.9
Comp. Ex. 1	PBR-4/PP-2 (75/25)	100	1.7	3.23/1.02	2.6
Ex. 4	PBR-1/PP-2 (50/50)	118	0.6	0.29/0.29	0.1
Ex. 5	PBR-2/PP-2 (50/50)	108	0.5	0.34/0.30	0.2
Ex. 6	PBR-3/PP-2 (50/50)	95	0.8	0.27/0.27	0.4
Comp. Ex. 2	PBR-4/PP-2 (50/50)	105	2.5	0.37/0.32	0.8

PBR ... Propylene-1-butene random copolymer

The properties of the propylene composite films obtained in the following examples were measured in the manner described below.

(1) Haze

Measured in accordance with ASTM D1003.

(2) Change of haze with time

The composite film was kept at 80 °C for 3 days and allowed to stand for cooling. Then, the haze was measured in the same manner as in the measurement (1).

(3) Glossiness

Measured in accordance with ASTM D523.

(4) Tear strength (Elmendorf tear strength)
Measured in accordance with ASTM D1922.

(5) Slip properties

Measured in accordance with ASTM D1894.

(6) Change of slip properties with time

The composite film was kept at 40 °C for one week and allowed to stand for cooling. Then, the slip properties

PP ... Polypropylene

^{...} Static friction coefficient/kinetic friction coefficient

100 Parts by weight of a mixture consisting of 25 % by weight of pellets of a random polypropylene having an isotactic index (I.I.) of 96 %, a melt index (230 °C) of 6.5 and a melting point of 140 °C and 75 % by weight of pellets of the propylene-1-butene random copolymer (B) (PBR) prepared above was mixed with 0.1 part by weight of an anti-blocking agent and 0.16 part by weight of a slip agent. The resulting mixture, i.e., the polypropylene composition (II), was melted in an extruder and then extruded on the substrate at 200 °C to form a laminated sheet.

The laminated sheet was stretched at 170 °C in a stretch ratio of 10 times in the lateral direction and then cooled with substantially maintaining the stretched state.

Thus, a biaxially stretched composite film consisting of a layer of the homopolypropylene (I) (substrate) having a thickness of 22 µm and a layer of the polypropylene composition (II) (heat-sealing layer) having a thickness of 3 µm was obtained. The results are set forth in Table 4.

Comparative Example 3

A composite film was obtained in the same manner as in Example 7 except that in the preparation of a polypropylene composition, the propylene-1-butene random copolymer obtained in Comparative Preparation Example 8 was used in place of the propylene-1-butene random copolymer (B) obtained in Preparation Example 5. The results are set to the in Table 4.

Examples 10 - 12

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A homopolypropylene (I) having an isotactic index (I.I.) of 96 % and a melt index of 6.5 was melted and fed to a composite film-forming die at a resin temperature of 240 °C.

Separately, in a different extruder, 100 parts by weight of a mixture consisting of 25 % by weight of pellets of a random polypropylene having a melt index (230 °C) of 6.5 and a melting point of 140 °C and 75 % by weight of pellets of the propylene-1-butene random copolymer (B) (PBR) prepared above was mixed with 0.1 part by weight of an anti-blocking agent and 0.16 part by weight of a slip agent, and they were melted in the extruder. The resulting polypropylene composition (II) was fed to the above-mentioned composite film-forming die at a resin temperature of 200 °C.

The homopolypropylene (I) and the polypropylene composition (II) were simultaneously extruded through the composite film-forming die, to obtain a composite film consisting of a layer of the homopolypropylene (I) (substrate) having a thickness of 45 µm and a layer of the polypropylene composition (II) (heat-sealing layer) having a thickness of 5 µm. The results are set torth in Table 4.

Comparative Example 4

A composite film was obtained in the same manner as in Example 10 except that the propylene-1-butene random copolymer obtained in Comparative Preparation Example 8 was used in place of the propylene-1-butene random copolymer (B) obtained in Preparation Example 5. The results are set forth in Table 4.

Table 4-I

		<u> </u>			
5	<u> </u>	Ex. 7	Comp.Ex.3	Ex. 8	Ex. 9
	(B) PBR	PBR-5 (Prep.Ex.5)	PBR-8 (Comp.Prep.Ex.8)	PBR-6 (Prep.Ex.6)	PBR-7 (Prep.Ex.7)
10	(1) Propylene content (mol%)	76	77	81	85
	1-Butene content (mol%)	24	23	19	15
15	(2) Intrinsic viscosity (η) (d//g)	1.63	1.91	1.65	1.67
	(3) Hw/Mn	2.05	3.40	2.10	2.18
20	(4) B value	1.00	0.92	1.02	1.03
	(5) Tm (°C)	. 92	110	103	110
	(6) Crystallinity (%)	40	48	49	53
25	(A) Polypro- pylene (wt1)	25	25	25	25
	sition (B) PBR (wt%)	75	75	75	75
30	(I) Substrate Thick- (µm)	22	22	22	22
35	ness (21)Heat- sealing layer (μm)	3	3	3	3
	(1) Haze (%)	1.4	1.7	0.6	0.8
	(2) Change of haze with time (%)	1.1	2.9	1.0	1.0
40	(3) Glossiness (%)	115	112	128	126
	(4) Slip properties /static_friction_	1_0	3.2	0.3	0.3
45	/kinetic friction	0,5	1.0	0.3	0.3
	(5) Slip properties with time				
50	/static friction /kinetic friction	0.6	0.7	0.3	0.3
	L				

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Table 4-I(continued)

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5		Ex. 7	Comp.Ex.3	Ex. 8	Ex. 9
	(B) PBR	PBR-5 (Prep.Ex.5)	PBR-8 (Comp.Prep.Ex.8)	PBR-6 (Prep.Ex.6)	PBR-7 (Prep.Ex.7)
10	(6) Blocking resistance(g/cm)	1.9	2.6	0.2	0.1
	(7) Blocking resistance with time (g/cm)	3.0	7.0	0.0	0.0
15	(8) Heat-sealing strength (g/15 mm) 85°C 90°C	40 480	20 50		,
20	95°C 100°C 105°C 110°C 115°C	630 550 730 520 520	130 480 570 570 520	90 300 540 440 440	20 320 540
25	120°C			470	500
30	(9) Hot tack properties (mm) 95°C 100°C 105°C 110°C 110°C	300 32 15 8 7	300 230 120 63 27	300 120 25	300 120
35	120°C 125°C 130°C 135°C 140°C	8 7 6 7 8 7	13 18 16 15	10 8 6 6	42 12 5 6 5

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Table 4-II

			,		
5		Ex. 10	Comp.Ex.4	Ex. 11	Ex. 12
	(B) PBR	PBR-5 (Prep.Ex.5)	PBR+8 (Comp.Prep.Ex.8)	PBR-6 (Prep.Ex.6)	PBR-7 (Prep.Ex.7)
10	(1) Propylene content	76	77	81	85
	(mol%) 1-Butene content (mol%)	24	23	19	15
15	(2) Intrinsic viscosity (η) (dl/g)	1.63	1.91	1.65	1.67
	(3) Mw/Mn .	2.05	3.40	2.10	2.18
20	(4) B value	1.00	0.92	1.02	. 1.03
	(5) Tm (°C)	92	110	103	110
	(6) Crystallinity (%)	40	48	49	53
25	(A) Polypro- pylene (wt%)	25	25	25	25
	pition (B) PBR (wt%)	75	75	75	75
30	(I) Substrate Thick- (μm)	45	45	45	45
35	ness (II)Heat- sealing layer (μm)	5 .	5	5	5
	(1) Haze (%)	1.8	2.1	4.2	2.1
	(2) Change of haze with time (%)	2.0	10.2	4.0	2.1
40	(3) Glossiness (%)	118	114	98	114
	(4) Slip properties /static friction	0-2	not_slipped	0.2_	0.2
45	/kinetic friction		not slipped		0.1
	(5) Slip properties with time /static friction				
50	/kinetic friction	ų.			

Comparative Example 5

A composite film was obtained in the same manner as in Example 13 except that in the preparation of a polypropylene composition, the propylene-1-butene random copolymer obtained in Comparative Preparation Example 8 was used in place of the propylene-1-butene random copolymer (B) obtained in Preparation Example 5. The results are set forth in Table 5.

Examples 15 - 17

A homopolypropylene (I) having an isotactic index (I.1.) of 96 % and a melt index of 6.5 was melted and fed to a composite film-forming die at a resin temperature of 240 °C.

Separately, in a different extruder, 100 parts by weight of a mixture consisting of 75 % by weight of pellets of a random polypropylene having a meit index (230 °C) of 6.5 and a meiting point of 140 °C and 25 % by weight of pellets of the propylene-1-butene random copolymer (B) (PBR) prepared above was mixed with 0.1 part by weight of an anti-blocking agent and 0.16 part by weight of a slip agent, and they were melted in the extruder. The resulting polypropylene composition (II) was ted to the above-mentioned composite film-forming die at a resin temperature of 200 °C.

The homopolypropylene (I) and the polypropylene composition (II) were simultaneously extruded through the composite film-forming die, to obtain a composite film consisting of a layer of the homopolypropylene (I) (substrate) having a thickness of 45 μ m and a layer of the polypropylene composition (II) (heat-sealing layer) having a thickness of 5 μ m. The results are set forth in Table 5.

Comparative Example 6

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A composite film was obtained in the same manner as in Example 15 except that the propylene-1-butene random copolymer obtained in Comparative Preparation Example 8 was used in place of the propylene-1-butene random copolymer (B) obtained in Preparation Example 5. The results are set forth in Table 5.

5

Table 5-I

		Еж. 13	Comp.Ex.5	Ex. 14
(E	B) PBR	PBR-5 (Prep.Ex.5)	PBR-8 (Comp.Prep.Ex.8)	PBR-6 (Prep.Ex.6)
	opylene content	76	יר	81
1-	ol%) Butene content ol%)	24	23	19
vi	trinsic scosity (dl/g)	1.63	1.91	1.65
(3) Mv	/Mn·	2.05	3.40	2.10
(4) B	value	1.00	0.92	1.02
(5) In	n (°C)	92	110	103
(6) Cı	ystallinity (%)	40	48	49
Compo-	(A) Polypro- pylene (wt%)	75	75	75
	(B) PBR (wt 1)	25	25	25
Thick-	(1) Substrate (μπ)	22	22	22
ness	(II)Heat- sealing layer (µm)	3	3	3
(1) H	aze (%)	1.7	1.5	1.9
	hange of haze ith time (%)	1.4	1.2	1.6
(3) G	lossiness (%)	122	123	122
	lip properties static friction	0.3	0.4	0.3
/	kinetic friction	0.3	C.4	0.3
	locking tendency g/cm)	0.1	0.3	0.1
	locking tendency ith time (g/cm)	0.3	0.6	0.3

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Table 5-I (Continued)

	Ex. 13	Comp.Ex.5	Ex. 14
(B) PBR	PBR-5 (Prep.Ex.5)	PBR-8 (Comp.Prep.Ex.8)	PBR-6 (Prep.Ex.6)
(7) Heat-sealing		1	
strength	1		
(g/15 mm)		1	
100°C	40		
105°C	100	50	
110°C	400	140	
115°C	510	350	30
120°C	540	460	180
125°C	500	460	510
130°C	460	460	480
135°C		470	480
140°C	{	480	480
150°C			470

Table 5-II

5		Ex. 15	Comp.Ex.6	Ex. 16	Ex. 17 .
	(B) PBR	PBR-5 (Prep.Ex.5)	PBR-8 (Comp.Prep.Ex.8)	PBR-6 (Prep.Ex.6)	PBR-7 (Prep.Ex.7)
10	(1) Propylene content	76	77	81	85
	(mol%) 1-Butene content (mol%)	24	23	19	15
15	(2) Intrinsic viscosity (η) (dl/g)	1.63	. 1.91	1.65	1.67
	(3) Mw/Mn	2.05	3.40	2.10	2.18
20	(4) B value	1.00	0.92	1.02	1.03
	(5) Tm (°C)	92	110	103	110
	(6) Crystallinity (%)	40	48	49	53
25	(A) Polypro- pylene (wt%)	75	75	75	75
	Sition (B) PBR (wt%)	25	25	25	25
30	(I) Substrate Thick- (µm)	45	45	45	45
35	(II) Heat- sealing layer (µm)	5	5	5	5
	(1) Haze (%)	2.4	2.5	2.0	2.1
40	(2) Change of haze with time (%)	2.0	2.8	2.0	2.4
	(3) Glossiness (%)	111	115	117	115
- 45	(4) Slip properties /static friction	0.2	0.3	0.3	0.2
	/kinetic friction	0.2	0.2	0.2	0.2
50	(5) Slip properties with time /static friction /kinetic friction	0.2	0.3	0.2	0.3
	1	1	I		

Table 5-II (Continued)

·	Ex. 15	Comp.Ex.6	Ex. 16	Ex. 17
(B) PBR	PBR-5 (Prep.Ex.5)	PBR-8 (Comp.Prep.Ex.8)	PBR-6 (Prep.Ex.6)	PBR-7 (Prep.Ex.7)
(6) Blocking resistance (g/cm)	0.2	0.3	0.2	0.3
(7) Blocking resistance with time (g/cm)	1.0	1.5	0.3	0.3
(8) Heat-sealing strength (g/15 mm) 100°C 110°C 120°C 130°C 140°C 150°C 160°C	10 160 1900 2130 2090 2400	10 80 1360 2380 2370 2520 2360	20 100 1130 2130 2090 2400	20 80 900 2100 1900 1890

Claims

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- 1. A polypropylene composition comprising:
 - (A) polypropylene in an amount of 5 to 95 % by weight, and
 - (B) a propylene-1-butene random copolymer in an amount of 95 to 5 % by weight; said propylene-1-butene random copolymer (B) having the following properties:
 - (1) the copolymer contains 50 to 95 % by mot of constituent units from propylene and 50 to 5 % by mot of constituent units from 1-butene;
 - (2) the intrinsic viscosity, as measured in decahydronaphthalene at 135 °C, is in the range of 0.1 to 12 dVg;
 - (3) the molecular weight distribution (Mw/Mn), as determined by get permeation chromatography (GPC), is
 - (4) the parameter B value indicating randomness of the copolymerized monomer sequence distribution is in.
- A composition according to claim 1, wherein the polypropylene (A) is a random copolymer of propylene and an 45 olefin other than propylene and the polypropylene (A) contains not more than 10 % by mol of units from the olefin other than propylene.
 - 3. A composition according to claim 1 or 2 wherein the propylene-1-butene random copolymer (B) further has the
 - (5) the melting point Tm, as measured by a differential scanning calorimeter, is in the range of 60 to 140 °C. and the melting point Tm and the 1-butene constituent unit content M (% by mol) satisfy the following relation:

$$-2.6M + 130 \le Tm - 2.3M + 155$$
; and

(6) the crystallinity C, as measured by X-ray diffractometry, and the 1-butene constituent unit content M (% by mol) satisfy the following relation:

C ≥ -1.5M + 75.

- A composition according to claim 1, 2 or 3, wherein the parameter B value of the propylene-1-butene random copolymer (B) is in the range of 1.0 to 1.3.
- A composition according to any one of the preceding claims, wherein the propylene-1-butene random copolymer
 (B) is obtainable using an otelin polymenzation catalyst comprising:
 - (a) a transition metal compound represented by the following formula (1):

$$\begin{array}{c|cccc}
X^1 & X^2 \\
M & R^3 & R^2 & R^3 \\
\hline
R^4 & R^4 & R^4
\end{array}$$
(1)

wherein M is a transition metal of Group IVa, Va or VIa of the periodic table,

R1 and R2 may be the same or different and are each a hydrogen atom, a halogen atom, a hydrocarbon group of 1 to 20 carbon atoms, a halogenated hydrocarbon group of 1 to 20 carbon atoms, a silicon-containing group, an oxygen-containing group, a sultur-containing group, a nitrogen-containing group or a phosphorus-containing group,

 ${\sf R}^3$ is a hydrocarbon group of 1 to 20 carbon atoms which may be substituted with a halogen atom or a siliconcontaining group,

R4 is a hydrogen atom or an alkyl group of 1 to 20 carbon atoms,

 X^1 and X^2 may be the same or different and are each a hydrogen atom, a halogen atom, a hydrocarbon group of 1 to 20 atoms, a halogenated hydrocarbon group of 1 to 20 carbon atoms, an oxygen-containing group or a sulfur-containing group, and

Y is a divalent hydrocarbon group of 1 to 20 carbon atoms, a divalent halogenated hydrocarbon group of 1 to 20 carbon atoms, a divalent silicon-containing group, a divalent germanium-containing group, a divalent tincontaining group, -O-, -CO-, -S-, -SO-, -SO₂-, NR5-, -P(R5)- -P(O) (R5)-, -BR5- or -AIR5- (R5 is a hydrogen atom, a halogen atom, a hydrocarbon group of 1 to 20 carbon atoms or a halogenated hydrocarbon group of 1 to 20 carbon atoms);

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(b-1) an organoaluminium oxy-compound, and/or

(b-2) a compound capable of reacting with the transition metal compound (a) to form an ion pair, and, optionally,

- _(c) an organoaluminium compound, ····
- 6. A composition according to claim 5, wherein R1 is methyl.
- 7. A composition according to claim 5, wherein R1 is a hydrocarbon group of 2 to 6 carbon atoms.
- 8. A polypropylene composite film comprising:
 - (I) a crystalline polypropylene layer, and
 - (II) a layer formed from the polypropylene composition as claimed in any one of the preceding claims which tayer is laminated on at least one surface of the crystalline polypropylene layer (I).
- 9. A film according to claim 8, wherein the polypropylene composition layer (II) comprises:



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